

**Variation in platinum group mineral and base metal sulfide assemblages  
in the Lower Group chromitites of the western Bushveld Complex,  
South Africa**

Bachmann, K.; Osbahr, I.; Tolosana-Delgado, R.; Chetty, D.; Gutzmer, J.;

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1 **Variation in platinum group mineral and base metal sulfide assemblages in the Lower Group**  
2 **chromitites of the western Bushveld Complex, South Africa**

3 Running Title: PGM and BMS in the LG chromitites, western Bushveld

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5 Kai Bachmann<sup>1,2\*</sup>, Inga Osbahr<sup>1,3</sup>, Raimon Tolosana-Delgado<sup>1</sup>, Deshenthree Chetty<sup>4</sup>, Jens Gutzmer<sup>1,2</sup>

6 <sup>1</sup>Helmholtz-Zentrum Dresden-Rossendorf, Helmholtz Institute Freiberg for Resource Technology,

7 Bautzner Landstraße 400, 01328 Dresden, Germany

8 <sup>2</sup>Department of Mineralogy, TU Bergakademie Freiberg, Brennhausgasse 14, D-09596

9 Freiberg/Sachsen, Germany

10 <sup>3</sup>Deutsche Rohstoffagentur (DERA) at the Bundesanstalt für Geowissenschaften und Rohstoffe,

11 Wilhelmstrasse 25-30, 13593 Berlin-Spandau, Germany

12 <sup>4</sup>Mineralogy Division, Mintek, Malibongwe Drive, 2125 Randburg, South Africa

13 \*Tel: +49 351 260 4426

14 Fax: +49 351 260 4440

15 k.bachmann@hzdr.de

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22 **Abstract**

23 The Lower Group chromitites of the Bushveld Igneous Complex are mined for chromite as a primary  
24 product. The recovery of platinum group elements and base metals (Ni, Cu) as by-products has the  
25 potential to add value to the chromite resources. This study focuses on the LG-6 and LG-6A  
26 chromitite seams at the Thaba mine located on the western limb of the Bushveld Complex. Platinum  
27 group minerals and base-metal sulfides are studied by mineral liberation analysis and electron  
28 microprobe analysis to define distinct assemblages and to evaluate the potential for beneficiation.  
29 Based on the results two distinct major mineral assemblages are defined: The first assemblage is rich  
30 in platinum group element-sulfides, along with variable proportions of malanite/cuprorhodsitite and  
31 alloys of Fe and Sn. The associated base metal sulfides are dominated by chalcopyrite and  
32 pentlandite, along with pyrite and subordinate millerite/violarite. Associated silicates are mainly  
33 primary magmatic orthopyroxene and plagioclase. The second assemblage is rich in platinum group  
34 element-sulfarsenides and -arsenides as well as -antimonides and -bismuthides, which are associated  
35 with a base metal sulfide assemblage dominated by pentlandite and Co-rich pentlandite. The  
36 assemblage is also marked by an abundance of alteration minerals, such as talc, serpentine and/or  
37 carbonates, which are closely associated with the platinum group minerals. Statistical evaluation  
38 reveals that these two mineral assemblages cannot be attributed to their derivation from different  
39 chromitite seams, but document the effects of pervasive hydrothermal alteration. Alteration  
40 evidently had similar effects on the different chromitite seams. The occurrence and distribution of  
41 the two characteristic assemblages has important implications for beneficiation. Assemblages rich in  
42 platinum group element-sulfides associated with base metal sulfides respond well to flotation,  
43 different to alteration assemblages rich in arsenides, antimonides and bismuthides. The nature of the  
44 gangue minerals will also impact platinum group mineral recovery as high phyllosilicate abundances,  
45 such as that encountered in the alteration assemblage may cause problems during flotation and lead  
46 to poor recoveries.

47 **Keywords:** SEM-based image analysis, ANOVA, Cluster Analysis, PGM, Thaba Mine, Bushveld  
48 Complex, EPMA

49 **Introduction**

50 The Bushveld Igneous Complex in South Africa is the world's largest layered mafic-ultramafic  
51 intrusion. The Rustenburg Layered Suite (RLS), comprising the most significant portion of the  
52 Bushveld Complex, consists of 7.5 to 9 km thick mafic and ultramafic cumulate rocks, covering an  
53 area of approximately 65,000 km<sup>2</sup>. The RLS was emplaced approximately 2,056 Ma ago  
54 (2,055.91±0.26; Zeh et al. 2015) and can be subdivided into the Marginal, Lower, Critical, Main, and  
55 Upper Zones (Hall 1932, Figure 1A).

56 The RLS hosts 21 chromitite seams (*e.g.*, Fourie 1959; Naldrett et al. 2012; 16 are shown in  
57 Figure 1B). According to their stratigraphic position these are subdivided into lower, middle and  
58 upper group chromitites. The Lower Group (LG) and Middle Group (MG) chromitites are extensively  
59 mined for chromite as a primary product (*e.g.*, the mines Doornbosch, Winterveld, Lannex,  
60 Tweefontein and Dwarsrivier; DERA 2013). The uppermost of the chromitite seams (UG-2), in  
61 contrast, is exploited for platinum-group elements (PGE) as the primary economic product as it  
62 reaches PGE concentrations up to 10 ppm  $\Sigma$ [PGE+Au] (Von Gruenewaldt et al. 1986; Lee 1996;  
63 Cawthorn 2011). Base metals and chromite are recovered from the UG-2 as by-products. Whilst all  
64 other chromitite seams do also contain elevated PGE concentrations, these range only from 0.5 to  
65 3 ppm (Wagner 1929; Von Gruenewaldt 1977; Von Gruenewaldt et al. 1986; Lee and Parry 1988;  
66 Cawthorn 1999; Barnes and Maier 2002; Naldrett et al. 2009a; Naldrett et al. 2012). Additionally, the  
67 chromitites display a progressive increase in PPGE (Pt+Pd+Rh) from the LG to the UG chromitites,  
68 whereas the content of IPGE (Os+Ir+Ru) remains broadly constant or increases only slightly (Naldrett  
69 and von Gruenewaldt 1989; Scoon and Teigler 1994; Naldrett et al. 2009b). This trend coincides with  
70 decreasing Cr/Fe ratios of the chromitites and results in a focus of mining the MG and LG seams for  
71 chromite (DERA 2013). The exploitation of PGE in the MG and LG seams is currently regarded as sub-  
72 economic (Oberthür et al., 2016).

73 Published knowledge on the mineralogy and deportment of PGE and base metals in the LG  
74 chromitite seams is scant (Teigler and Eales 1993; Scoon and Teigler 1994; Naldrett et al. 2009b,  
75 2012; Junge et al. 2016; Oberthür et al. 2016) and beside the investigation of Maier and Barnes  
76 (Maier and Barnes 1999, and references therein) at Union Section, only scarce information is  
77 available in particular for the northern part of the western limb of the Bushveld Complex. The UG-2,  
78 in contrast, has been investigated in great detail (*e.g.*, McLaren and De Villiers 1982; Gain 1985;  
79 Hiemstra 1985, 1986; Maier and Barnes 2008; Cawthorn 2011; Junge et al. 2014, 2015; Osbahr et al.  
80 2014). Junge et al (2016) and Oberthür et al. (2016) emphasized the similarity of PGM of the LG-6  
81 with the UG-2 chromitite. However, with samples derived for these studies from reworked chromite  
82 dumps, they may yield representative results but lack any geological context. Given the scarcity of  
83 published data, it is not surprising that the mechanism of PGE enrichment in chromitite seams  
84 remains a subject of debate. Controversial aspects include the sulfide-poor nature of the RLS (S  
85 contents commonly < 100 ppm; Scoon and Teigler 1994), and the presence of discrete PGM  
86 containing almost the entire budget of PGE (*e.g.*, Hiemstra 1986).

87 Studies on the beneficiation characteristics of PGM and base metal sulfides (BMS) in Bushveld  
88 chromitite seams other than the UG-2 are very scant – as most published contributions focus on  
89 those resources currently exploited (UG-2, Platreef and Merensky Reef) (*e.g.*, Penberthy et al. 2000;  
90 Bulatovic 2003; Shackleton et al. 2007a,b; Brough et al. 2010; Becker et al. 2008; Chetty et al. 2009,  
91 Voordouw et al. 2010, Bushell 2012, Smith et al. 2013, Becker et al. 2014). These studies confirm that  
92 characteristics such as quantitative mineralogy, deportment, mineral association and grain sizes play  
93 a major role in PGM beneficiation. The type of PGM species, their liberation and association are  
94 particularly relevant for floatability (*e.g.*, Penberthy et al. 2000; Chetty et al 2009) as well as the  
95 association of PGM with BMS (Xiao and Laplante, 2004). PGM association with gangue minerals  
96 displays a separate set of challenges for flotation processes (*e.g.*, Chetty et al. 2009). Furthermore,  
97 Penberthy et al. (2000) reported that the grain size of PGM also affects floatability, with very fine (*i.e.*  
98 <3 µm) PGM considered as slow floaters.

99 Despite these constraints a possible recovery of base metals and PGE as by-product during  
100 exploitation of chromite would certainly add value and increase resource efficiency of chromite  
101 mining ventures. A successful example in this regard is Sylvania Platinum Ltd., reworking chromite  
102 dumps of the LG-6 and MG-1/2 chromitites from the eastern and western Bushveld Complex and  
103 producing saleable concentrates of both PGE and chromite (Junge et al. 2016; Oberthür et al. 2016).

104 This study focuses on the LG-6 and LG-6A chromitite seams located at the Thaba mine, a chromite  
105 mine situated on the western limb of the Bushveld Complex (Figure 1A) that is operated by Cronimet  
106 Chrome Mining SA (Pty) Ltd., a subsidiary of the Cronimet Mining Group. The proven reserves at  
107 Thaba total 23.6 Mt of chromite ore at 43.6 wt% Cr<sub>2</sub>O<sub>3</sub> resulting in ca. 10 Mt contained Cr<sub>2</sub>O<sub>3</sub> (DERA  
108 2013). Mining started in 2011 and the life of mine is estimated at 26 years (DERA 2013). The mine  
109 exploits the LG-6/LG-6A as well as the MG-1 to MG-4A chromitite seams. Any PGE recovery needs to  
110 follow, or be integrated with, the recovery of chromite as the primary product.

111 In the present study, samples were obtained from three drill core intersections of the LG-6 and LG-6A  
112 chromitites at Thaba mine. The aims of the study are:

- 113 (i) to characterize the PGM and base metal sulfide and Fe-sulfide (BMS) assemblages by *in-*  
114 *situ* analysis of polished thin section surfaces of LG-6/LG-6A chromitite using scanning  
115 electron microscopy–energy dispersive X-ray spectrometry (SEM-EDS)-based image  
116 analysis techniques, complemented by electron microprobe analysis (EPMA),
- 117 (ii) to provide data on the composition of major silicate minerals and chromite,
- 118 (iii) to relate observed systematic changes in mineralogy and mineral association to  
119 geological processes,
- 120 (iv) to discuss the obtained data with respect to published work and to assess implications  
121 for the beneficiation of the chromitite ores.

122 The results are used to constrain the nature, distribution and origin of PGM and BMS in the  
123 chromitites of the RLS. Furthermore, our study provides important constraints for eventual  
124 beneficiation, as we consider not only qualitative mineralogy and mineral assemblage, but also  
125 quantify relevant attributes such as mineralogy, mineral grain sizes and mineral association for PGM  
126 and BMS. The *in-situ* methodology approach used in this study is preferred compared to the more  
127 widely employed method of studying grain mounts of PGM concentrates (*e.g.*, Junge et al. 2016;  
128 Oberthür et al. 2016) as it eliminates limitations to produce statistically significant results, for  
129 parameters such as mineral association, as shown elsewhere (*e.g.* Penberthy et al. 2000; Voordouw  
130 et al. 2010; Viljoen et al. 2012; Smith et al. 2013).

### 131 **Analytical Methods**

132 Access to diamond drill core intersections of LG-6 and LG-6A chromitite seams at Thaba mine was  
133 granted by Cronimet Mining Group. Drill cores were logged to document the lithological architecture  
134 of each intersection. Quarter drill cores were sampled with the goal to sample “pristine ores” below  
135 the extent of present-day weathering, *i.e.* to depth in excess of 50m below the present-day land  
136 surface, across the mine lease area. This approach was successful as none of the studied samples  
137 revealed any evidence of supergene oxidation of BMS. Six different drill core locations were selected  
138 for this study (Table 1), three of them for sampling of LG-6 (EL28, ZK149, ZK144) and three for LG-6A  
139 (EL32, ZK136, SC42). Deflections were considered wherever available. Samples from deflections are  
140 always demarcated by adding a D to the drill hole location name (Table 1). Chemical assay data for  
141 some of the selected drill cores were made available by Cronimet Mining Group. Available assay data  
142 refer always to entire chromitite seams; no assay data were available for drill cores ZK144, EL32D,  
143 and ZK136D.

144 The general sampling approach follows that of Voordouw et al. (2010). To investigate, if the LG-6 is  
145 bottom- or top-loaded with respect to PGM and BMS, each drill core intersection was divided into  
146 five distinct sectors: sector 1 covers the bottom 10%, sector 3 the mid 40% and sector 5 the top 10%

147 of the seam (sector 2 and 4 are each 20 %). Each sector was sampled randomly to prepare an  
148 individual polished thin section for study.. A similar approach was applied to the sampling of the LG-  
149 6A, but samples of sector 1 and 5 were set onto the contact between host rock and chromitite  
150 wherever this was possible. This resulted in a higher overall silicate proportion in the modal  
151 mineralogy than for the LG6 samples. Each section was cut vertical to stratigraphy.

#### 152 *Mineral Liberation Analysis (MLA)*

153 Thirty polished thin sections of ca. 150  $\mu\text{m}$  in thickness were prepared at the Helmholtz Institute  
154 Freiberg for Resource Technology (HIF). Each of these sections was analyzed twice by MLA on two  
155 polished surfaces well-separated by re-grinding and re-polishing. These two surfaces are referred to  
156 as surface A and surface B; all data and detailed investigations on the reproducibility of the MLA  
157 measurements can be found in the electronic supplementary material (Appendix A, B and C). The  
158 sections were analyzed twice to increase the number of PGM and BMS grains identified and to  
159 constrain intra-sample variation. Furthermore, two sample surfaces were analyzed as in-run  
160 duplicates to check for internal consistency of data acquisition.

161 The MLA instrument used in this study is located at the HIF and comprises a scanning electron  
162 microscope FEI Quanta 650F equipped with two Bruker Quantax X-Flash 5030 energy-dispersive X-ray  
163 spectrometers and the MLA 3.1.4 software suite for automated data acquisition. Consistent  
164 operating conditions were applied (Table 2). DataView (Fandrich et al. 2007) software was used for  
165 further processing of the data. All samples were analyzed using the sparse phase liberation (SPL\_Lt)  
166 and the grain X-ray mapping (GXMAP) mode (Fandrich et al., 2007). Standard spectra were collected  
167 for all relevant minerals. Selected grains of silicates, BMS and PGM were then analyzed by electron  
168 microprobe (EPMA, see below). More detailed information about the functionality of a MLA and the  
169 joint offline processing of EPMA and MLA data can be found in Gu (2003) and Bachmann et al.  
170 (2017).



171 As shown by Voordouw et al. (2010), abundances of individual PGM, sulfides and silicates may differ  
172 from each other by an order of magnitude; the proportions of individual minerals are thus described  
173 within chemically similar groups rather than in absolute terms (e.g., area%<sup>PGM</sup> refers to the area% of  
174 individual PGM among all PGM).

#### 175 *Electron probe microanalysis (EPMA)*

176 Electron probe microanalysis was performed with a JEOL JXA 8530F at the HIF equipped with a field  
177 emission electron gun and five wavelength dispersive spectrometers (WDS). Five different analyzer  
178 crystals TAP, PETJ, PETL, PETH, and LIFH on five spectrometers were utilized to measure chemical  
179 compositions of various base metal and iron sulfides (BMS) as well as PGM. Quantitative EPMA  
180 analyses were also performed on major silicates and chromite. Detailed information about the  
181 methodology and the entire data set is provided in the electronic supplementary (Appendix A and D).

182 The assignment of the analyzed elements to detector crystals and spectrometers as well as peak and  
183 background positions, dwell times and the lower limit of detection during PGM and BMS analyses are  
184 shown in Table 3. All quantitative analyses were performed with a focused beam at an accelerating  
185 voltage of 12 kV/ 20 kV and a beam current of 100 nA/ 30 nA. Analyses were corrected according to  
186 the measurement protocol of Osbahr et al. (2015), slightly modified by a time-resolved offline  
187 overlap correction and a step-wise overlap correction of multiple interferences. EPMA results were  
188 further used to generate mineral standard spectra with a known composition for MLA  
189 measurements.

#### 190 *Statistics*

191 Statistical data evaluation and comparison of properties across the data set was obtained via  
192 ANalysis Of VAriance (ANOVA, e.g. Fahrmeir and Hammerle, 1984), which explains the variance of  
193 several target (predicted) variables by the influence of certain explanatory factors. In our case the  
194 modal mineralogy in respectively area%<sup>PGM</sup>, area%<sup>BMS</sup>, area%<sup>Gangue</sup> (Gangue: alteration silicates,  
195 silicates, carbonates) are predicted variables. Other minerals occurring in trace quantities, such as

196 monazite, zircon, etc, were not considered as they add no relevant information. Chromite was not  
197 considered because the chromite content is strongly dependent on the sampling location, *i.e.*  
198 massive chromitite, chromitite with silicate intercalations or chromitite – host rock contact.  
199 Explanatory factors considered were: The section surface (A/B), the relative stratigraphic position  
200 within a seam “intra seam” (top, top-middle, middle, bottom-middle and bottom), and either the  
201 seam (LG-6 vs. LG-6A) or the borehole (EL28D-LG-6, ZK149-LG-6, ZK144-LG-6, EL32D-LG-6A, ZK136D-  
202 LG-6A vs.SC42-LG-6A, “inter-core” effects). ANOVA allows to test in an organized manner, which of  
203 these variables exert a significant, systematic control on the area% composition of the mineral  
204 groups considered. This is achieved by F-tests targeting each explanatory variable, and checking the  
205 null hypothesis that the residual variability of a model including that target variable is the same as a  
206 model without it. If this would be the case, then the target variable would not reduce the uncertainty  
207 about the explained composition, and should be discarded. In practical terms, for those variables for  
208 which the F-test returned a significance level of  $p < 0.05$ , the null hypothesis is rejected, hence the  
209 target variable is accepted as having a significant influence on the composition. Significance levels  
210 considered are: 0-0.001 highly significant, 0.001-0.01 moderately significant, 0.01-0.05 significant,  
211 0.05-1 not significant.

212 A cluster analysis for the mineral association of PGM and BMS was performed to support the  
213 separation into distinct ore types suitable for further treatment in a mineral beneficiation process.  
214 The R software environment (R Core Team 2016) with the additional packages “mclust” (Fraley and  
215 Raftery 2002) and “compositions” (van den Boogaart et al. 2014) were used for data analysis. The  
216 algorithm “mclust” requires a choice of the characteristics (orientation, size and shape) of the  
217 covariance matrix which are allowed to vary between groups: the analysis was done in “VVV”-mode,  
218 meaning that each cluster was allowed to have a totally different covariance matrix (Fraley et al.  
219 2012). All modal mineralogy subcompositions were treated with the statistical framework of  
220 compositional data analysis (Aitchison, 1986; Tolosana-Delgado, 2012), to avoid the effects of

221 spurious correlation occurring in such closed subcompositions (Chayes, 1960). To get rid of these  
222 effects, the solution chosen here is to work with ratios or log-ratios between components.

## 223 **Results**

224 Results are presented in three sections. The first provides a brief description of the lithological  
225 architecture and qualitative petrographic appearance of the LG-6 and LG-6A chromitite seams  
226 sampled. The following section reports the results of EMP analysis, *e.g.* the mineral chemistry of  
227 PGM and BMS. This data is complemented by analyses of silicates and chromite provided in the  
228 electronic supplement (Appendix A). Finally, MLA results are evaluated in terms of modal mineralogy  
229 (sub-compositions PGM/BMS/Gangue as defined above) and mineral association (PGM/BMS).

### 230 *Lithology and Mineralogy*

231 This section provides largely qualitative descriptions of the studied chromitite seams – and their  
232 mineralogy. Quantitative mineralogical data is provided later in this contribution. The LG-6 and LG-6A  
233 chromitite seams at Thaba mine are well developed as massive, single seams that are usually hosted  
234 by pyroxenite. The LG-6 seam as the primary mining target averages 0.85 m in thickness, varying  
235 between 0.34 m and 1.37 m; the Cr<sub>2</sub>O<sub>3</sub> content averages 43.1 % and the Cr:Fe ratio is 1.58. The LG-6A  
236 occurs stratigraphically approximately 6 m above the LG-6 seam and averages at 0.23 m in thickness,  
237 containing on average 41.2 % Cr<sub>2</sub>O<sub>3</sub> with a Cr:Fe ratio of 1.48. Locally, cm-thin chromitite stringers  
238 (such as in sample EL028-LG-6) or pegmatoidal pyroxenite (ZK136-LG-6A and EL32D-LG-6A) occur in  
239 the pyroxenite host rock. Footwall (0.2 m) and hanging wall (1 m) of EL32D-LG-6A consist of  
240 serpentinite (altered pyroxenite) while some of the pyroxenite in intersection ZK149-LG-6 is also  
241 strongly serpentized.

242 All chromitite samples selected for this study are best described as massive chromitites that are  
243 comprised mostly of chromite, orthopyroxene and plagioclase. Thin intercalations of silicate rocks as  
244 well as up to a few cm long silicate oikocrysts are locally common in these massive chromitites.

245 Chromite grains are subspherical in shape, with average diameters of < 0.5 mm (max. ~1 mm), and  
246 smooth grain margins. Chromite grains may host inclusions of silicates, sulfides and Fe-Ti oxides.  
247 Subvertical veinlets are common, containing variable amounts of mica, amphibole and minor  
248 carbonates. Especially in sample EL32D-LG-6A, chromite grains appear fractured and are cemented  
249 by a younger generation of chromite.

250 The silicate minerals have been subdivided into two assemblages. The first assemblage includes  
251 silicates regarded as orthomagmatic in origin (called silicates group), the other assemblage includes  
252 all (hydrous) silicates that are considered to have formed by post-magmatic hydrothermal alteration  
253 (called alteration silicates group). The silicates group is dominated by pyroxene (more orthopyroxene  
254 than clinopyroxene), followed by feldspar, and traces of olivine, while quartz is almost absent. Mica is  
255 present in very minor abundances and is biotitic in composition. Feldspar compositions are  
256 dominated by plagioclase, alkali feldspar occurs only in trace amounts. In intersections ZK149-LG-6  
257 and EL32D-LG-6A, feldspar is entirely absent.

258 The alteration silicates group is dominated by amphibole, chlorite (mainly clinochlor) and talc, while  
259 only minor serpentine is present. Carbonate minerals (mainly of dolomitic compositions) were also  
260 identified as part of the hydrothermal alteration assemblage and occur only in very minor amounts.  
261 A notable exception is the occurrence of dolomite in chromitite seam LG-6A of drill core EL32D.

262 PGM and BMS occur in all samples in trace amounts. PGM are observed as monomineralic grains,  
263 which may show internal zonation, but also as aggregates of variable complexity (for representative  
264 examples see Figures 2B-H). Equivalent circle diameters (ECDs, value of the diameter of a circle  
265 having the same area as the measured grain/particle; *e.g.*, Fisher et al. 1987) of PGM grains/  
266 aggregates typically range between <1 and 20 µm, ECDs of up to 35 µm are scant. Average ECDs are  
267 consistently between 4 µm and 5 µm. PGM are often locked or attached to BMS but also occur  
268 enclosed in or interstitial to chromite, silicates and alteration silicates. PGM enclosed within chromite  
269 grains consist mainly – but not exclusively - of laurite (Figure 2G). Members of the laurite [RuS<sub>2</sub>] -

270 erlichmannite [OsS<sub>2</sub>] series are also the main carriers of IPGE (iridium-subgroup; Ir, Ru, Os). Minerals  
271 belonging to the cooperite-braggite-vysotskite [PtS – (Pt,Pd)S – PdS] series and members of the  
272 thiospinel solid solution series malanite [CuPt<sub>2</sub>S<sub>4</sub>]-cuprorhodsites [CuRh<sub>2</sub>S<sub>4</sub>] are major carriers of the  
273 PPGE (platinum subgroup; Pt, Pd, Rh). Sperrylite [PtAs<sub>2</sub>], PGE-antimonides, -bismuthides and -  
274 bismuthotellurides such as geversite [PtSb<sub>2</sub>], and stibiopalladinite [Pd<sub>5+x</sub>Sb<sub>2+x</sub>] with minor sudburyite  
275 [PdSb], insizwaite [PtBi<sub>2</sub>], and traces of genkinite [(Pt,Pd)<sub>4</sub>Sb<sub>3</sub>] and moncheite [(Pt,Pd)(Te,Bi)<sub>2</sub>] are of  
276 lesser abundance as are PGE-alloys including rustenburgite [Pt<sub>3</sub>Sn] and tetraferroplatinum [PtFe].  
277 Traces of native platinum [Pt] were also identified. Sulfarsenides (platarsite [PtAsS], hollingworthite  
278 [RhAsS] and irarsite [IrAsS]) with variable compositions contribute significantly to the overall PGE  
279 budget in some samples.

280 BMS usually occur as aggregates up to 300 μm in diameter or as single grains (<100 μm), interstitial  
281 to chromite or enclosed in silicates. Smaller (up to 100 μm) aggregates of BMS were found as  
282 inclusions in chromite. Importantly, pyrrhotite is almost absent. In general, BMS are dominated by  
283 pentlandite [(Ni,Fe)<sub>9</sub>S<sub>8</sub>], which in some cases contains minor concentrations of Co. Other important  
284 Ni-sulfides are millerite [NiS] and violarite [Fe<sup>2+</sup>Ni<sub>2</sub><sup>3+</sup>S<sub>4</sub>]. Violarite often rims and replaces pentlandite  
285 (Figure 2A, D). Some examples of millerite and violarite do also contain minor amounts of Co.  
286 Chalcopyrite and pyrite only occur in some samples in significant amounts – relative to the other  
287 BMS. Galena, stibnite and sphalerite were detected only in negligible abundance – they are not  
288 considered further.

### 289 *Geochemistry and Mineral Chemistry*

290 5PGE+Au (Pt,Pd,Rh,Ir,Ru,Au) contents are around 1 ppm for LG-6 samples and 1.5 ppm for the LG-6A,  
291 with a Pt/Pd ratio around 2.2 and 4 and Pt/Ru ratios around 0.9 for intersections ZK144-LG-6 and  
292 ZK149-LG-6, respectively. In general, the total sulfur content ranges between 100 – 200 ppm  
293 (<250 ppm), except intersection ZK144-LG-6, which contains only around 70 ppm. Arsenic contents

294 are usually below the detection limit (<0.5 ppm), except in a few samples of ZK149-LG-6 (up to  
295 9 ppm). Detailed data is provided with the electronic supplement (Appendix A and E).

296 Mineral chemistry data was obtained for all relevant rock-forming mineral groups, except chlorite  
297 and mica. Pyroxenes, amphiboles, serpentine and talc, as well as chromite compositions are well  
298 discriminated by their mineral chemistry. Two different types of pyroxenes were detected, an  
299 orthopyroxene (hereafter referred to as enstatite with  $Mg/(Mg+Fe^{2+})$  ratios ranging from 0.82-0.84)  
300 and a clinopyroxene with augitic to diopsidic composition (hereafter referred to as diopside).  
301 Amphiboles can be best described as magnesio-hornblende. Feldspars yield plagioclase compositions  
302 (Median =  $An_{66}$ ). In general, chromite displays  $Mg/(Mg+Fe^{2+})$  cation ratios around 0.42 (sample  
303 EL32D-LG-6A around 0.33), while  $Cr/(Cr+Al)$  cation ratios plot at 0.73 (EL32D-LG-6A:  $\sim 0.64$ ). Detailed  
304 data and methodology are provided in the electronic supplement (Appendix A and D).

305 The composition of PGM and BMS is collated in Tables 4 and 5. Mineral chemistry data were used to  
306 provide proper identification of PGM in MLA data sets – and to characterize BMS in greater detail.  
307 Pentlandite displays rather variable compositions, especially regarding the concentration of Co. For  
308 the purpose of this study we refer to pentlandite containing > 2 wt% Co as Co-rich pentlandite. Pyrite  
309 analyses reveal Co and Ni contents up to 1.5 wt% and 3.5 wt%, respectively.

310 A notable number of pyrite, pentlandite, Co-rich pentlandite and violarite analyses as well as a few  
311 chalcopyrite analyses yielded measureable PGE concentrations. Pyrite, pentlandite and Co-rich  
312 pentlandite yield concentrations up to 2-3 wt% Rh, Pt, Ru and Ir, around 1 wt% Pd and up to 0.5 wt%  
313 Os. Chalcopyrite was found to contain PGE contents below 1 wt% and violarite up to 0.4 wt% Rh and  
314 0.2 wt% Pd, respectively. Despite the fact that several previous studies have suggested that BMS can  
315 contain significant amounts of PGE (Osbahr et al. 2013; Osbahr et al. 2014; Junge et al. 2014), the  
316 concentrations reported in this study are unusually high. We tentatively consider the PGE as being  
317 localized as submicroscopic inclusions of discrete PGM in the BMS (see Figure 2H), occasionally hit by  
318 spot analyses during the EPMA measurements.

319 *Quantitative Mineralogy and Microfabric Data*

320 Quantitative mineralogical and microfabric data were obtained by MLA analysis. In order to facilitate  
321 comparison, the rock-forming minerals were grouped as follows (sorted by abundance): Chromite  
322 (Al-rich spinel, various chromite compositions), silicates (ortho- and clinopyroxene, feldspar, olivine,  
323 quartz, biotite), alteration silicates (amphiboles, chlorite, muscovite, serpentine, talc), carbonates  
324 (calcite, dolomite), others (*e.g.*, apatite, barite, monazite, rutile, titanite, zircon). For PGM and BMS  
325 compositional sub-groups were defined (Table 6). The relative abundance of these is expressed as  
326  $\text{area}\%^{\text{PGM}}$  and  $\text{area}\%^{\text{BMS}}$  – this was done to ease comparison between different samples. The same  
327 was done for gangue mineral groups (alteration silicates, silicates, carbonates) – these are expressed  
328 as  $\text{area}\%^{\text{Gangue}}$ .

329 The quantitative mineralogy of all sampled seams is given in Table 7. For this purpose, data obtained  
330 for all five samples collected from each seam was combined. Data for PGM and BMS for every thin  
331 section surface analyzed is provided in the electronic supplement (Appendix C). This investigation  
332 uses relative abundances of different PGM subgroups for classification. As illustrated in Figure 3A,  
333 minerals belonging to the (Ru,Ir,Os)<sub>2</sub>S<sub>2</sub> subgroup are present in all samples, but comprise rather  
334 different proportions of the total PGM population (LG-6: 17 up to 59 total  $\text{area}\%^{\text{PGM}}$ ; LG-6A: 6 up to  
335 64 total  $\text{area}\%^{\text{PGM}}$ ). In samples EL28D-LG-6, ZK144-LG-6, ZK136D-LG-6A and SC42-LG-6A (Pt,Pd)S and  
336 (Pt,Rh)<sub>2</sub>CuS<sub>4</sub> are common PPGE carriers subordinate only to alloys containing Fe and Sn; the latter  
337 range from 30 to 68 total  $\text{area}\%^{\text{PGM}}$  in the LG-6 and from 33 to 65 total  $\text{area}\%^{\text{PGM}}$  in the LG-6A seam.  
338 (Pt,Rh)<sub>2</sub>CuS<sub>4</sub> is the dominant PPGE carrier in ZK144-LG-6. In contrast, the PGM assemblage in samples  
339 ZK149-LG-6 and EL32D-LG-6A is dominated by (PGE)As,AsS and (PGE)Sb,Bi, comprising 44 to 76 total  
340  $\text{area}\%^{\text{PGM}}$  in ZK149-LG-6 and 53 to 85 total  $\text{area}\%^{\text{PGM}}$  in EL32D-LG-6A.

341 The BMS mineralogy is dominated by Co-rich pentlandite, chalcopyrite, millerite, violarite and pyrite  
342 as well as traces of pyrrhotite (Figure 3B). The sulfide proportions are highly variable. In samples  
343 ZK149-LG-6 and EL32D-LG-6A Co-rich pentlandite represents the majority of the BMS, while

344 chalcopyrite, pyrite as well as minor millerite dominate sample ZK144. Samples EL28D, ZK136D and  
345 SC42 contain variable proportions of Co-rich pentlandite, chalcopyrite, pyrite/ (pyrrhotite), violarite  
346 and minor millerite.

347 Figure 3C displays the relative distribution of PGM and BMS of intersections of the LG-6 and LG-6A  
348 seams. While ZK144-LG-6 has a rather low BMS content and an assemblage dominated by  
349 chalcopyrite and minor millerite, both Co-rich pentlandite dominated intersections (ZK149-LG-6 and  
350 EL32D-LG-6A) display a high sulfide content (resulting in a distinctly higher sulfur content, *i.e.* sample  
351 ZK149-LG-6 compared to sample ZK144-LG-6). In contrast, the PGM contents are rather similar, with  
352 somewhat lower abundances in EL28D-LG-6 and EL32D-LG-6A and distinctly higher PGM contents in  
353 sample ZK144-LG-6.

#### 354 **Statistical Assessment**

355 The application of statistical methods is needed to deal with the large number of samples and the  
356 multivariate datasets. This leads to the differentiation of three substantially distinct PGM-BMS  
357 assemblages:

- 358 (I) PGM-sulfides and alloys of Fe and Sn  $\pm$  PGM-Cu-sulfides occur together with a BMS  
359 assemblage dominated by pentlandite + chalcopyrite + pyrite
- 360 (II) PGM-Cu-sulfides  $\pm$  PGM-sulfides  $\pm$  alloys of Fe and Sn; corresponding BMS are dominated  
361 by chalcopyrite + pyrite + millerite  $\pm$  pentlandite
- 362 (III) PGM-sulfarsenides and PGM-arsenides occur together with PGM-antimonides and -  
363 bismuthides and scant tellurides; corresponding BMS are strongly dominated by  
364 pentlandite and Co-rich pentlandite  $\pm$  chalcopyrite  $\pm$  pyrite.

365 The following sections document the relative contribution of ANOVA and cluster analysis to this  
366 result.

367



## 368 ANOVA

369 An ANOVA was performed – followed by standard F-tests – to investigate the variability of the  
370 samples and to determine assemblages for classification according to the sub-compositions  
371  $\text{area}\%^{\text{PGM}}$ ,  $\text{area}\%^{\text{BMS}}$  and  $\text{area}\%^{\text{Gangue}}$ . Detailed results can be found in the electronic supplementary  
372 material (Appendix A). Results suggest that the “nugget effect” (small scale variability) is negligible  
373 for all sub-compositions. The variance between samples of a different stratigraphic position within a  
374 sampled seam, *i.e.* if seams have systematically different mineral assemblages within their respective  
375 confines, should be noted – but remains small. Furthermore, there are no systematic changes in  
376 mineral abundances observed between the two chromitite seams LG-6 and LG-6A. Thus, the  
377 variability between certain drill cores explains up to 80 % of the total variability – depending on the  
378 selected sub-composition. A detailed investigation of the ANOVA model showed a strong negative  
379 correlation of PGE -sulfarsenides and PGE-antimonides, -bismuthides vs. PGE –sulfides, -Cu-sulfides  
380 and -alloys of Fe,Sn (for  $\text{area}\%^{\text{PGM}}$ ), along with (Co-rich) pentlandite against the other four BMS  
381 considered, namely chalcopyrite, millerite, pyrite and violarite (for  $\text{area}\%^{\text{BMS}}$ ). Furthermore, a  
382 negative correlation of PGE-Cu-sulfides vs. PGE –sulfides and –alloys of Fe,Sn along with chalcopyrite  
383 and millerite against pyrite and violarite was observed. The  $\text{area}\%^{\text{Gangue}}$  assemblages are best  
384 classified by separating alteration silicates and silicates-rich from carbonate-rich drill cores, followed  
385 by a separation of alteration silicates from silicate-rich cores. While carbonate-rich samples are  
386 associated with PGM-BMS assemblage (III), silicate-rich samples dominate PGM-BMS assemblage (I).  
387 Significant amounts of alteration silicates may occur in samples belonging to PGM-BMS assemblages  
388 (II) and (III).

## 389 *Cluster analysis*

390 For a quantitative assessment of the variability of the mineral association a cluster analysis was  
391 performed according to the sub-compositions of PGM (without IPGE sulfides) and BMS. The results  
392 were linked to the mineral assemblages defined in the previous chapter (ANOVA). Detailed  
393 information can be found in the electronic supplementary material (Appendix A). Despite the very low

394 total abundance of BMS (< 0.02 area% in all samples), PGM show a strong preferred association with  
395 BMS; they also show a close association with alteration silicates, occurring both in interstitial  
396 positions and as inclusion. PGM are only to a much lesser extent associated with silicates and  
397 chromite. PGM occur predominantly interstitial to chromite (with minor inclusions cf. Figure 2G).  
398 BMS show a very similar association to PGM.

## 399 **Discussion**

400 Appropriate statistical assessment of quantitative mineralogical data suggests the presence of three  
401 distinct PGM-BMS mineral assemblages in the LG-6 and LG-6A chromitite seams at Thaba mine. The  
402 LG-6 and LG-6A intersections show striking differences between different drill cores, resulting in  
403 variable PGM, BMS, and gangue mineral assemblages. The (para)-genetic relevance of the  
404 documented differences between these three assemblages is evaluated below. Finally, implications  
405 for recovery of PGM and BMS are briefly discussed.

### 406 *General Aspects*

407 The LG-6/LG-6A chromitites at Thaba mine share numerous similarities to lateral equivalents at  
408 other locations in the Bushveld Complex (e.g., Teigler and Eales 1993; Eales et al. 1993; Scoon and  
409 Teigler 1994; Naldrett et al. 2009b, 2012). Such similarities include:

- 410 (1) chromitites are developed as single seams hosted by pyroxenite,
- 411 (2) Cr:Fe ratio is 1.58 for the LG-6 and 1.48 for the LG-6A, respectively,
- 412 (3) orthopyroxenes yield  $Mg/(Mg+Fe^{2+})$  ratios ranging from 0.82-0.84,
- 413 (4) intercumulus feldspar compositions within pyroxenites in the Lower Critical Zone are those of  
414 labradorite (Median =  $An_{66}$ ),
- 415 (5) limited variation in chromite compositions,
- 416 (6) uniform PGE grades (5E + Au) around 1 ppm (LG-6) and 1.5 ppm (LG-6A),
- 417 (7) Pt/Ru ratios around 0.9, Pt/Pd ratios ranging from 2 to 4, and low  $S_{tot}$  ranging from 70 to  
418 200 ppm.

419 We thus deem it likely that the observations made here can be transferred to other occurrences of  
420 LG chromitites within the Bushveld Complex.

#### 421 *Mineral assemblages*

422 Different to previous investigations the present study offers quantitative mineralogical and  
423 microfabric data on *in-situ samples* (*i.e.* polished thin samples not modified by comminution or  
424 mineral separation) for PGM and BMS in the LG-6/LG-6A chromitites of the Bushveld Complex. The  
425 data provide unique insight into different PGM, BMS and rock-forming mineral assemblages and  
426 associations.

427 PGM assemblages are dominated by various Pt-Pd-Rh sulfides (cooperite-braggite and malanite-  
428 cuprorhodsite), laurite (the main carrier of the IPGE), sulfarsenides, sperrylite and Pt-Fe alloys; PGE-  
429 tellurides and PGE-bismuthotellurides are largely absent. These observations are rather similar to  
430 previous studies (*e.g.*, Junge et al 2016; Oberthür et al. 2016) carried out on mineral concentrates.

431 BMS occur only in very minor abundances. Pentlandite, Co-bearing pentlandite, chalcopyrite and  
432 pyrite dominate, whilst millerite is much less common. The virtual absence of pyrrhotite is noted. The  
433 BMS mineralogy is thus rather similar to that reported in previous studies for Bushveld chromitites  
434 (*e.g.*, Junge et al 2016; Oberthür et al. 2016).

435 Yet, despite these broad similarities, LG-6 and LG-6A chromitite intersections at Thaba mine reveal  
436 significant and systematic mineralogical variations that remained previously unnoticed. These  
437 mineralogical variations are observed in both the LG-6 and LG-6A – and can thus not be used to  
438 distinguish the two seams from one another. Rather, the differences need to be described as lateral  
439 variations independent of stratigraphic position.

440

441

442 Three different PGM-BMS assemblages were recognized (Figure 4):

- 443 (I) PGM-sulfides and alloys of Fe and Sn  $\pm$  PGM-Cu-sulfides occur together with a BMS  
444 assemblage dominated by pentlandite + chalcopyrite + pyrite
- 445 (II) PGM-Cu-sulfides  $\pm$  PGM-sulfides  $\pm$  alloys of Fe and Sn; corresponding BMS are dominated  
446 by chalcopyrite + pyrite + millerite  $\pm$  pentlandite
- 447 (III) PGM-sulfarsenides and PGM-arsenides occur together with PGE-antimonides, -  
448 bismuthides and scant -bismuthotellurides; corresponding BMS are strongly dominated  
449 by pentlandite and Co-rich pentlandite  $\pm$  chalcopyrite  $\pm$  pyrite

450 All three of these PGM-BMS mineral assemblages are interpreted to be the product of postmagmatic  
451 alteration, variably modifying an orthomagmatic assemblage.

452 Mineral assemblage (I) can be regarded as the most pristine assemblage, revealing a low alteration  
453 silicate to orthomagmatic silicate ratio. A relatively high BMS concentration and a variable PGM and  
454 BMS mineral association mark this assemblage. BMS display variable abundances of pentlandite,  
455 chalcopyrite, pyrite and minor violarite. PGE carriers are PGE sulfides and PGE-Cu sulfides with  
456 variable proportions of PGE alloys of Fe and Sn. This is especially evident in intersection SC42-LG-6A,  
457 which contains significant amounts of violarite. While traces of millerite in EL28D-LG-6 and SC42-LG-  
458 6A as well as significant amounts of PGE alloys of Fe and Sn are consistent with beginning  
459 desulfidization, violarite is typically regarded as a product of late stage hydrothermal alteration of  
460 mixtures of iron-nickel sulfides (Dunn and Howes 1996; Tenailleau et al. 2006). Based on a statistical  
461 analysis a "typical" mineral association for PGM in assemblage (I) was defined. The dominating PGE-  
462 sulfides and -alloys of Fe and Sn are strongly associated to BMS and show only moderate affinity to  
463 silicates and alteration silicates (Figure 5).

464 Assemblage (II), best typified by sample ZK144-LG-6, is marked by the abundance of alteration  
465 silicates, although the alteration silicate to silicate ratio remains moderate. The total BMS content in  
466 assemblage (II) is rather low, as is the total sulfur content ( $\sim$ 70 ppm). The BMS assemblage is

467 dominated by chalcopyrite, pyrite and millerite. Minerals of the malanite-cuprorhodsite solution  
468 series are by far the most important carriers for PPGE in this assemblage.

469 BMS and PGM present in assemblage (II) suggest an origin by removal of Fe and S by either late  
470 magmatic fluids and/or by reaction of sulfide with chromite from an orthomagmatic assemblage (e.g.  
471 Naldrett et al 2012 and references therein). The high amounts of Ni-rich sulfides are explained  
472 through re-equilibration of Fe from magmatic sulfide liquid with chromite (Naldrett and Lehmann  
473 1988).

474 The high abundance of hydrous silicates associated with assemblage (II) suggests that corrosive  
475 action of hydrothermal fluids further modified the BMS assemblage. That these fluids may play a  
476 major role to explain the loss of sulfides was shown, for example, for the UG-2 (Penberthy and  
477 Merkle 1999; Li et al. 2004, Voordouw et al., 2010). High malanite-group PGM concentrations in  
478 chromitite were recently documented by Oberthür et al. (2016). The latter authors proposed that  
479 part of the orthomagmatic chalcopyrite experienced a similar fate as the re-equilibrated Fe-sulfides  
480 and has reacted to some extent with cooperite/braggite to form malanite. Although the PGM  
481 mineralogy displays some similarities with assemblage (I), such as a similar PGM spectrum, the  
482 mineral association of these PGM is distinctly modified and strongly dominated by alteration  
483 silicates, minor silicates and lack of association with BMS.

484 Mineral assemblage (III) is best represented by intersections ZK149-LG-6 and EL32D-LG-6A and can  
485 be further linked to gangue mineral assemblages. According to the ANOVA results, the gangue  
486 mineral assemblage is closely related to assemblage (II). However, assemblage (III) generally displays  
487 a significantly higher ratio in alteration minerals to orthomagmatic silicates; in drill core ZK149-LG-6  
488 this is due to an abundance of alteration silicates, whereas in drill core EL32D-LG-6A carbonates are  
489 particularly widespread. Host rocks of these two intersections display strong serpentinization and  
490 thus provide further evidence for pervasive hydrothermal alteration. Furthermore, chromite analyses  
491 display somewhat higher  $Mg/(Mg+Fe^{2+})$  and  $Cr/(Cr+Al)$  cation ratios (in sample EL32D-LG-6A). A

492 model including alteration by As-bearing fluids is favored over crystallization directly from As-bearing  
493 melt as proposed by, e.g., Gervilla et al. (1996). Reasons for this interpretation include: (1)  
494 Investigated samples have very low As contents and contain no other arsenides, such as nickeline,  
495 etc., which would be expected in As-bearing melts. (2) PGE (sulf-)-arsenide-rich drill core  
496 intersections display other prominent alteration features, as described above. (3) PGE arsenide-rich  
497 assemblages of secondary origin are widely reported for the Bushveld Complex (Peyerl 1982; Kinloch  
498 1982, Voordouw et al 2010). These arguments also support a secondary origin for PGE-antimonides  
499 and -bismuthides. The PGE-(sulf)-arsenides, -antimonides and -bismuthides occur closely associated  
500 with alteration silicates, carbonates or are interstitial to chromite. They show only a moderate  
501 affinity to BMS and are almost absent as inclusions in chromite (Figure 5). Beside the carbonate-  
502 dominated samples, the statistical assessment yielded similar results for the PGM mineral association  
503 in assemblage (III) and assemblage (II), due to the strong affinity to alteration silicates. Nevertheless,  
504 the low PGM affinity to silicates but higher affinity interstitial to chromite and to BMS, mainly  
505 pentlandite, points to different processes responsible for the modification of the original  
506 orthomagmatic mineral assemblage.

507 Laurite-group minerals, as the main IPGE carrier, deserve specific mention. They behave very  
508 different to other PGM, and show stable proportions over all drill core intersections. According to the  
509 statistical analysis laurite-group minerals remain unaffected by alteration processes. The same is true  
510 for pyrite and pyrrhotite. Even though these two Fe-sulfides were reliably separated from  
511 assemblage (III) in the ANOVA model, a further preferential association to assemblages (I) or (II) was  
512 not possible to elucidate. This might be caused by the variable appearance of pyrite, as mineral of  
513 primary (cf. Figure 2C) or possibly secondary origin (cf. Figure 2A, D), for example by replacing  
514 pyrrhotite or as a by-product of the violarite alteration of pentlandite.

515

516

517 *Implications for beneficiation*

518 Given the fact that PGM and BMS were investigated *in-situ* in this study, the following relevant  
519 parameters may be extracted and quantified from the data set for the potential recovery of PGM and  
520 BMS: 1. Mineral species; 2. Mineral association; 3. Grain size distribution and 4. Gangue mineralogy  
521 (*e.g.* Penberthy et al. 2000; Chetty et al. 2009). All these parameters are known to have a significant  
522 impact on PGM recovery by flotation (*e.g.* Becker et al. 2013; Ndlovu et al. 2014).

523 Table 8 summarizes all relevant parameters of the three noted assemblages for mineral  
524 beneficiation; similarities and differences between these assemblages are highlighted. Studies on the  
525 floatability of different PGM species in the UG-2 chromitite (*e.g.* Penberthy et al. 2000; Chetty et al.  
526 2009) suggest a higher floatability of sulfide-rich PGM assemblages compared to sulfarsenide-rich  
527 assemblages. Flotation performance is also influenced by BMS and gangue mineralogy, as well as the  
528 textural relations between PGM and associated minerals.

529 Relatively high recoveries may thus be expected for the BMS-rich assemblage (I), since the PGM  
530 assemblage is not only sulfide-rich, but typically intergrown with a BMS assemblage marked by an  
531 abundance of fast-floating chalcopyrite (Penberthy et al. 2000; Smith et al. 2013). In contrast, the  
532 BMS content of assemblage (II) is significantly lower and PGM show rather little association to the  
533 contained BMS. Therefore, recoveries of PGM may be expected to be somewhat lower, despite the  
534 presence of PGM sulfides and chalcopyrite. Assemblage (III) can be expected to yield even lower  
535 recoveries. This is due to the dominance of slow-floating sulfarsenides and a close association of  
536 PGM with alteration silicates, chromite and carbonates – minerals that either do not float  
537 (chromite/carbonate) or that need to be depressed in the given flotation process (*e.g.*, talc). The high  
538 content of phyllosilicates in the alteration silicate association of assemblage (III) may be expected to  
539 have rheological impacts in flotation, leading to even lower recoveries, as discussed by *e.g.*, Becker et  
540 al. (2013) and Ndlovu et al. (2014).

541 In addition to the mineralogy of the three assemblages, the grain sizes of PGM and BMS will play an  
542 important role in defining recoveries during flotation. The average grain size of 4-5  $\mu\text{m}$  ECD for the  
543 PGM and up to 300  $\mu\text{m}$  ECD for BMS in all samples studied – irrespective of the actual assemblage.  
544 This suggests that PGM and BMS should be amenable to flotation, if fully liberated and not reduced  
545 further in size during comminution. PGM and BMS particle sizes  $<3 \mu\text{m}$  are generally considered to  
546 have poor flotation characteristics (Chetty et al. 2009). Nevertheless, to achieve sufficient liberation  
547 of the majority of the minute PGM grains will likely require fine milling – beyond the typical flotation  
548 feed of 45 – 80 % passing 75  $\mu\text{m}$  typically used. However, the effects of ultrafine milling on liberation  
549 characteristics of PGM and BMS mineral grains and grain aggregates are beyond the purpose of this  
550 contribution. If PGM are associated with BMS – either because of intergrowth or because of the  
551 presence of minute inclusions of PGM – the grain sizes of PGM may in fact not be relevant for  
552 flotation - but rather the aggregate sizes and liberation of the BMS. These aggregates are typically  
553 larger than 30  $\mu\text{m}$ , often up to a few 100  $\mu\text{m}$  which will easily float and further improve PGE  
554 recovery.

## 555 **Conclusion**

556 Based on a combination of mineralogical and micro-analytical data, complemented by a tailored  
557 statistical assessment, our study delivers the first systematic evaluation of PGM and BMS  
558 assemblages in the LG's of the Bushveld Complex. According to our assessment, no truly unaltered  
559 orthomagmatic mineral assemblage is preserved at the site of study – the LG-6 and LG-6A chromite  
560 seams exploited at Thaba mine of the northernmost part of the western limb of the Bushveld  
561 Complex. Alteration is documented by changes to the PGM, BMS and gangue mineralogy, while  
562 chromite remains almost unaffected. Alteration assemblages are similar in both of the two studied  
563 LG seams. It was further demonstrated that lateral variation between different drill cores dominates  
564 over vertical variation within seam as well as between the LG-6 and LG-6A seams.



565 A “typical” PGM spectrum for all Bushveld chromitites as proposed by *e.g.*, Junge et al. (2016) or  
566 Oberthür et al. (2016) is not supported by the results of the present study. Even though all included  
567 groups of the “typical” PGM were detected, different assemblages of different origin can be  
568 distinguished. This study thus extends the assemblages well-documented for the UG-2 (Kinloch 1982;  
569 Peyerl 1982; Penberthy and Merkle 1999; Voordouw et al. 2010) to the LG chromitites. Our results  
570 may be thus of general applicability, not only for the Thaba mine but also wider parts of the  
571 chromitites of the Bushveld Complex, taking the similarities between, *e.g.*, the general mineralogy,  
572 mineral chemistry, geology, host rocks within the Bushveld into account.

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- 749

750 **Tables**

751 Table 1 General information and available bulk chemical data of all investigated samples.

| Borehole      | Length      | From          | To            | Cr <sub>2</sub> O <sub>3</sub> | FeO         | SiO <sub>2</sub> | MgO        | Al <sub>2</sub> O <sub>3</sub> | CaO        | P         |
|---------------|-------------|---------------|---------------|--------------------------------|-------------|------------------|------------|--------------------------------|------------|-----------|
| <i>LG-6</i>   | <i>in m</i> | <i>in m</i>   | <i>in m</i>   | <i>all data in wt%</i>         |             |                  |            |                                |            | ppm       |
| EL28          | 0.23        | 167.04        | 167.27        | 43.3                           | 27.2        | 4.2              | 8.7        | 14.2                           | 0.4        | 20        |
| <b>EL28D</b>  | <b>0.30</b> | <b>167.45</b> | <b>167.75</b> | <b>43.5</b>                    | <b>28.1</b> | <b>3.1</b>       | <b>8.5</b> | <b>16.3</b>                    | <b>0.4</b> | <b>20</b> |
| <b>ZK149</b>  | <b>0.83</b> | <b>94.68</b>  | <b>95.51</b>  | <b>43.2</b>                    | <b>25.0</b> | <b>5.0</b>       | <b>9.2</b> | <b>14.0</b>                    | <b>0.3</b> | <b>20</b> |
| ZK149D        | 0.83        | 94.73         | 95.56         | 43.6                           | 25.4        | 4.5              | 9.7        | 14.5                           | 0.3        | 20        |
| <b>ZK144</b>  | <b>0.87</b> | <b>294.98</b> | <b>295.85</b> | -                              | -           | -                | -          | -                              | -          | -         |
| ZK144D        | 0.93        | 294.89        | 295.82        | 44.0                           | 26.9        | 3.5              | 8.7        | 13.6                           | 0.3        | 20        |
| <i>LG-6A</i>  | <i>in m</i> | <i>in m</i>   | <i>in m</i>   | <i>all data in wt%</i>         |             |                  |            |                                |            | ppm       |
| <b>EL32D</b>  | <b>0.19</b> | <b>252.59</b> | <b>252.78</b> | -                              | -           | -                | -          | -                              | -          | -         |
| ZK136         | 0.20        | 268.95        | 269.15        | 44.7                           | 25.7        | 4.1              | 11.3       | 14.6                           | 0.4        | 28        |
| <b>ZK136D</b> | <b>0.20</b> | <b>269.22</b> | <b>269.42</b> | -                              | -           | -                | -          | -                              | -          | -         |
| <b>SC42</b>   | <b>0.25</b> | <b>344.60</b> | <b>344.85</b> | <b>43.8</b>                    | <b>23.2</b> | <b>4.1</b>       | <b>9.7</b> | <b>14.2</b>                    | <b>0.2</b> | <b>25</b> |
| SC42D         | 0.24        | 344.58        | 344.82        | 42.9                           | 24.6        | 3.2              | 10.8       | 15.2                           | 0.3        | 65        |

752 Please note that chromitite intersections in bold are considered in detail in this study. Boreholes  
753 marked with "D" are deflections of the motherhole and regarded as their most similar sample.  
754 Analyses were performed at Set Point Laboratories, Johannesburg. Cr<sub>2</sub>O<sub>3</sub>, FeO, SiO<sub>2</sub>, MgO, Al<sub>2</sub>O<sub>3</sub>, CaO,  
755 P were analyzed by X-ray fluorescence (XRF). "-" = no geochemical analyses are available.

756

757 Table 2 Summary of MLA parameters.

| SPL parameters        |             | GXMAP parameters     |           |
|-----------------------|-------------|----------------------|-----------|
| Voltage (kV)          | 25          | Voltage (kV)         | 25        |
| Probe current (nA)    | 10          | Probe current (nA)   | 10        |
| HFW (μm)              | 750         | HFW (μm)             | 1000      |
| BSE calib.            | Au 252      | BSE calib.           | Au 250    |
| Resolution (pixels)   | 1000 × 1000 | Resolution (pixels)  | 500 × 500 |
| Pixel size (μm/px)    | 0.75        | Pixel size (μm/px)   | 2         |
| Quartz EDX-count      | 2000        | Quartz EDX-count     | 2000      |
| BSE Range             | 100 - 255   | GXMAP trigger        | 25        |
| Frame Guide Size (px) | 30          | Step size (px)       | 8 × 8     |
| Min. grain size (px)  | 2           | Min. grain size (px) | 4         |

758 SPL = sparse phase liberation measurement mode, HFW = horizontal field width, BSE = back scattered  
759 electrons, px = pixel, GXMAP = grain-based X-ray mapping measurement mode, EDX = energy  
760 dispersive X-ray spectroscopy

761

762

763 Table 3 Instrumental parameters applied to PGM/BMS analysis on the EPMA.

| Element/<br>Line  | Spectrometer/<br>Crystal | Peak<br>position<br>(mm) | Lower<br>Backgr.<br>(mm) | Upper<br>Backgr.<br>(mm) | Measurement<br>Time Peak (s) | Measurement<br>Time Backgr.<br>(s) | 12 kV Limit of<br>Quantification*<br>Sulfide/ PGM<br>(ppm) | 20 kV Limit of<br>Quantification*<br>Sulfide/ PGM<br>(ppm) | Standards (ASTIMEX<br>Standards Ltd.) |
|-------------------|--------------------------|--------------------------|--------------------------|--------------------------|------------------------------|------------------------------------|--|--|---------------------------------------|
| As L $\alpha$ 1,2 | 1 TAP                    | 105.14                   | 6.127                    | 6.563                    | 60                           | 15                                 | 213/ 297   | 330/ 480   | Arsenopyrite_AST                      |
| S K $\alpha$ 1    | 2 PETJ                   | 171.599                  | 5.061                    | -                        | 40                           | 20                                 | 117/ 170   | 157/ 177   | Pentlandite_AST                       |
| Pd L $\beta$ 1    | 3 PETL                   | 133.048                  | 4.970                    | 9.601                    | 40                           | 20                                 | 493/ 653   | 417/ 770   | Palladium_AST                         |
| Rh L $\alpha$ 1,2 | 3 PETL                   | 147.419                  | 5.086                    | 4.703                    | 40                           | 10                                 | 243/ 327   | 227/ 367   | Rhodium_AST                           |
| Au M $\alpha$ 1   | 3 PETL                   | 187.047                  | 10.418                   | 9.281                    | 40                           | 10                                 | 473/ 1023  | 603/ 817   | Gold_AST                              |
| Pt M $\alpha$ 1   | 3 PETL                   | 193.563                  | 10.028                   | 2.769                    | 40                           | 10                                 | 513/910  | 573/ 1050  | Platinum_AST                          |
| Ir M $\alpha$ 1   | 3 PETL                   | 200.458                  | 4.410                    | 3.024                    | 40                           | 10                                 | 536/970  | 597/ 1040  | Iridium_AST                           |
| Os M $\alpha$     | 3 PETL                   | 207.728                  | 4.093                    | 3.127                    | 40                           | 10                                 | 543/ 880   | 597/ 940   | Osmium_AST                            |
| Sb L $\beta$ 1    | 4 PETH                   | 103.058                  | 8.361                    | 9.804                    | 40                           | 10                                 | 467/ 763   | 450/ 620   | Stibnite_AST                          |
| Te L $\alpha$ 1   | 4 PETH                   | 105.095                  | 10.989                   | 7.960                    | 40                           | 10                                 | 210/ 330   | 197/ 270   | Tellurium_AST                         |
| Ru L $\alpha$ 1   | 4 PETH                   | 155.161                  | 3.218                    | -                        | 40                           | 20                                 | 133/ 210   | 143/203  | Ruthenium_AST                         |
| Bi M $\alpha$ 1   | 4 PETH                   | 163.863                  | 12.114                   | 13.374                   | 40                           | 10                                 | 337/ 550   | 430/ 560   | BismuthSelenide_AST                   |
| Mo L $\beta$ 1    | 4 PETH                   | 165.793                  | 14.082                   | 17.831                   | 40                           | 10                                 | 433/ 633   | 467/ 633   | Molybdenum_AST                        |
| Cu K $\alpha$ 1   | 5 LIFH                   | 107.259                  | 1.960                    | 2.131                    | 40                           | 10                                 | 867/ 1510  | 297/ 467   | Copper_AST                            |
| Ni K $\alpha$ 1   | 5 LIFH                   | 115.466                  | 6.521                    | 2.887                    | 40                           | 10                                 | 660/ 1130  | 257/ 427   | Pentlandite_AST                       |
| Co K $\alpha$ 1   | 5 LIFH                   | 124.555                  | 3.106                    | 2.745                    | 40                           | 10                                 | 473/ 787   | 240/ 343   | Cobalt_AST                            |
| Fe K $\alpha$ 1   | 5 LIFH                   | 134.797                  | 5.100                    | 3.000                    | 40                           | 10                                 | 383/ 633   | 197/ 313   | Pentlandite_AST                       |

764 All standards supplied by ASTIMEX Standards Ltd. Limit of Quantification = 10 × limit of detection.

765

Table 4 Representative EPMA analyses of major PGM. Z – numbers of atoms per formula unit. b.d.l. – below detection limit; apfu – atoms per formula unit.

| No.                              | 300         | 193         | 414          | 511          | 127        | 145        | 186        | 51         | 90          | 115         | 99           | 243         | 71              | 329             | 294         | 273         | 239               |   |
|----------------------------------|-------------|-------------|--------------|--------------|------------|------------|------------|------------|-------------|-------------|--------------|-------------|-----------------|-----------------|-------------|-------------|-------------------|---|
| Comment                          | EL28D-3_080 | EL28D-2_071 | ZK136D-2_054 | ZK136D-2_113 | SC42-2_097 | SC42-2_115 | SC42-2_156 | SC42-2_021 | ZK149-5_063 | ZK149-5_088 | ZK149-5_072c | EL32D-4_049 | ZK149-5_044c    | EL32D-4_135     | EL32D-4_131 | EL32D-4_079 | EL32D-4_045       |   |
| Seam                             | LG-6        | LG-6        | LG-6A        | LG-6A        | LG-6A      | LG-6A      | LG-6A      | LG-6A      | LG-6        | LG-6        | LG-6         | LG-6A       | LG-6            | LG-6A           | LG-6A       | LG-6A       | LG-6A             |   |
| Mineral                          | Cooperite   | Cooperite   | Braggite     | Vysotskite   | Vysotskite | Vysotskite | Malanite   | Laurite    | Laurite     | Laurite     | Platarsite   | Platarsite  | Hollingworthite | Hollingworthite | Sperrylite  | Geversite   | Stibiopalladinite |   |
| <i>all data in wt%</i>           |             |             |              |              |            |            |            |            |             |             |              |             |                 |                 |             |             |                   |   |
| As                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | bdl        | bdl        | bdl         | bdl         | 27.64        | 29.08       | 35.02           | 30.14           | 35.11       | 0.6         | bdl               |   |
| S                                | 15.37       | 14.98       | 18.8         | 22.71        | 23.65      | 24.91      | 26.29      | 38.57      | 38          | 36.69       | 11.75        | 13.14       | 10.91           | 15.99           | 0.34        | 0.09        | 0.24              |   |
| Pd                               | 0.43        | 0.59        | 26.46        | 32.71        | 46.6       | 63.25      | 0.1        | bdl        | bdl         | 0.06        | bdl          | 0.03        | bdl             | bdl             | 3.26        | bdl         | 66.9              |   |
| Rh                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | 10.57      | 0.68       | 0.24        | 0.9         | 4.03         | 16.3        | 24.88           | 44.95           | 0.06        | bdl         | bdl               |   |
| Au                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | bdl        | bdl        | bdl         | bdl         | bdl          | bdl         | bdl             | bdl             | bdl         | bdl         | 0.14              |   |
| Pt                               | 80.27       | 80.18       | 53.24        | 34.9         | 16.49      | 0.38       | 38.77      | bdl        | bdl         | bdl         | 35.45        | 22.07       | 24.06           | 0.36            | 48.59       | 42.33       | 0.06              |   |
| Ir                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | 8.28       | 2.65       | 2.8         | 4.91        | 1.72         | 2.52        | bdl             | 1.65            | bdl         | bdl         | 0.03              |   |
| Os                               | 0.1         | 0.05        | bdl          | bdl          | bdl        | bdl        | 0.5        | 1.72       | 6.68        | 8.8         | 2.02         | 5.56        | 0.32            | 0.24            | 0.03        | 0.06        | bdl               |   |
| Sb                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | bdl        | bdl        | bdl         | bdl         | bdl          | 0.43        | bdl             | 3.77            | 4.45        | 47.52       | 30.42             |   |
| Te                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | bdl        | bdl        | bdl         | bdl         | bdl          | bdl         | bdl             | bdl             | 0.02        | 0.25        | bdl               |   |
| Ru                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | 0.07       | 55.3       | 52.55       | 47.42       | 14.89        | 8.28        | 3.24            | 2.35            | bdl         | 0.01        | bdl               |   |
| Bi                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | bdl        | bdl        | bdl         | bdl         | bdl          | bdl         | bdl             | bdl             | 5.12        | 7.96        | bdl               |   |
| Mo                               | bdl         | bdl         | bdl          | bdl          | bdl        | bdl        | 1.2        | 0.15       | 0.15        | 0.12        | bdl          | bdl         | bdl             | bdl             | bdl         | bdl         | bdl               |   |
| Cu                               | 0.04        | 0.04        | 0.08         | 0.1          | 0.18       | bdl        | 12.06      | 0.02       | bdl         | bdl         | bdl          | bdl         | bdl             | bdl             | bdl         | 0.01        | bdl               |   |
| Ni                               | 2.39        | 1.77        | 2.46         | 10.55        | 11.01      | 10.81      | 0.4        | 0.07       | 0.03        | 0.14        | 0.14         | 0.4         | 0.21            | 0.1             | 0.19        | 0.04        | 0.27              |   |
| Co                               | 0.03        | 0.02        | bdl          | 0.02         | bdl        | bdl        | 1.48       | bdl        | bdl         | bdl         | bdl          | 0.07        | 0.02            | 0.02            | 0.06        | bdl         | 0.05              |   |
| Fe                               | 1.89        | 2.05        | 1.11         | 1.37         | 1.34       | 1.05       | 1.42       | 1.18       | 0.73        | 1.37        | 1.34         | 0.52        | 0.28            | 0.7             | 0.43        | 0.68        | 0.69              |   |
| Total                            | 100.52      | 99.67       | 102.15       | 102.37       | 99.26      | 100.41     | 101.14     | 100.35     | 101.18      | 100.41      | 98.98        | 98.37       | 98.92           | 100.26          | 97.66       | 99.55       | 98.79             |   |
| <i>all data is given as apfu</i> |             |             |              |              |            |            |            |            |             |             |              |             |                 |                 |             |             |                   |   |
| As                               | -           | -           | -            | -            | -          | -          | -          | -          | -           | -           | 0.963        | 0.959       | 1.154           | 0.851           | 1.688       | 0.036       | -                 |   |
| S                                | 0.987       | 0.982       | 1.002        | 1.011        | 1          | 0.986      | 4.018      | 2          | 2.006       | 1.993       | 0.956        | 1.012       | 0.84            | 1.055           | 0.038       | 0.012       | 0.058             |   |
| Pd                               | 0.008       | 0.012       | 0.425        | 0.439        | 0.594      | 0.754      | 0.004      | -          | -           | 0.001       | -            | 0.001       | -               | -               | 0.11        | -           | 4.863             |   |
| Rh                               | -           | -           | -            | -            | -          | -          | 0.503      | 0.011      | 0.004       | 0.015       | 0.102        | 0.391       | 0.597           | 0.924           | 0.002       | -           | -                 |   |
| Au                               | -           | -           | -            | -            | -          | -          | -          | -          | -           | -           | -            | -           | -               | -               | -           | -           | 0.006             |   |
| Pt                               | 0.847       | 0.864       | 0.466        | 0.255        | 0.115      | 0.002      | 0.974      | -          | -           | -           | 0.474        | 0.279       | 0.304           | 0.004           | 0.897       | 0.969       | 0.002             |   |
| Ir                               | -           | -           | -            | -            | -          | -          | 0.211      | 0.023      | 0.025       | 0.045       | 0.023        | 0.032       | -               | 0.018           | -           | -           | 0.001             |   |
| Os                               | 0.001       | 0.001       | -            | -            | -          | -          | 0.013      | 0.015      | 0.059       | 0.081       | 0.028        | 0.072       | 0.004           | 0.003           | 0.001       | 0.001       | -                 |   |
| Sb                               | -           | -           | -            | -            | -          | -          | -          | -          | -           | -           | -            | 0.009       | -               | 0.066           | 0.132       | 1.743       | 1.933             |   |
| Te                               | -           | -           | -            | -            | -          | -          | -          | -          | -           | -           | -            | -           | -               | -               | 0.001       | 0.009       | -                 |   |
| Ru                               | -           | -           | -            | -            | -          | -          | 0.003      | 0.91       | 0.88        | 0.817       | 0.384        | 0.202       | 0.079           | 0.049           | -           | 0.001       | -                 |   |
| Bi                               | -           | -           | -            | -            | -          | -          | -          | -          | -           | -           | -            | -           | -               | -               | 0.088       | 0.17        | -                 |   |
| Mo                               | -           | -           | -            | -            | -          | -          | 0.061      | 0.003      | 0.003       | 0.002       | -            | -           | -               | -               | -           | -           | -                 |   |
| Cu                               | 0.001       | 0.001       | 0.002        | 0.002        | 0.004      | -          | 0.93       | 0.001      | -           | -           | -            | -           | -               | -               | -           | 0.001       | -                 |   |
| Ni                               | 0.084       | 0.063       | 0.071        | 0.257        | 0.254      | 0.234      | 0.033      | 0.002      | 0.001       | 0.004       | 0.006        | 0.017       | 0.009           | 0.004           | 0.012       | 0.003       | 0.035             |   |
| Co                               | 0.001       | 0.001       | -            | 0.001        | -          | -          | 0.123      | -          | -           | -           | -            | 0.003       | 0.001           | 0.001           | 0.004       | -           | 0.006             |   |
| Fe                               | 0.07        | 0.077       | 0.034        | 0.035        | 0.033      | 0.024      | 0.125      | 0.035      | 0.022       | 0.043       | 0.063        | 0.023       | 0.012           | 0.027           | 0.028       | 0.055       | 0.095             |   |
| Z                                | 2           | 2           | 2            | 2            | 2          | 2          | 7          | 3          | 3           | 3           | 3            | 3           | 3               | 3               | 3           | 3           | 3                 | 7 |

Table 5 Representative compositions of major sulfide minerals (BMS) as analyzed by EPMA.

| Mineral            |        | S           | Cu          | Fe          | Co        | Ni          | Pb  |
|--------------------|--------|-------------|-------------|-------------|-----------|-------------|-----|
| (1) Chalcopyrite   | Median | 34.62       | 32.91       | 30.28       | bdl       | 0.02        | bdl |
|                    | Range  | 33.77-37.19 | 31.71-33.27 | 29.63-32.05 | bdl       | bdl-0.20    | bdl |
| (2) Pyrite         | Median | 53.32       | 0.02        | 45.01       | 0.61      | 0.33        | bdl |
|                    | Range  | 49.21-55.17 | bdl-0.37    | 41.34-46.52 | bdl-1.49  | 0.01-3.46   | bdl |
| (3) Pentlandite    | Median | 33.22       | bdl         | 27.68       | 0.91      | 37.57       | bdl |
|                    | Range  | 30.14-41.19 | bdl-0.15    | 19.12-31.71 | 0.08-1.14 | 32.95-40.62 | bdl |
| (4) Co-Pentlandite | Median | 33.52       | bdl         | 28.86       | 4.46      | 32.67       | bdl |
|                    | Range  | 30.90-35.57 | bdl         | 21.03-30.86 | 3.40-5.29 | 27.89-35.31 | bdl |
| (5) Millerite      | Median | 35.73       | bdl         | 2.56        | bdl       | 61.39       | bdl |
|                    | Range  | 35.13-37.78 | bdl         | 2.17-4.59   | bdl-0.80  | 57.86-62.18 | bdl |
| (6) Violarite      | Median | 42.84       | 0.08        | 29.30       | 0.29      | 23.82       | bdl |
|                    | Range  | 41-32-44.29 | bdl-0.15    | 27.66-30.91 | 0.25-0.33 | 23.31-27.26 | bdl |

Notes: (1): n=43; (2): n=44; (3): n=70; (4): n=35, pentlandite analyses with >2.00 wt% Co; (5): n=3; (6): n=4; b.d.l. – below detection limit.

Table 6 Grouping of PGM (left) and BMS (right), respectively.

| PGM                                   |  |                                | BMS   |                     |                                |
|---------------------------------------|--|--------------------------------|-------|---------------------|--------------------------------|
| Group                                 | Mineral  | Group <sub>Geostatistics</sub> | Group | Mineral             | Group <sub>Geostatistics</sub> |
| (PGE)AsS                              | hollingworthite, irarsite, platarsite, sperrylite    | PGM(1)                         | Ccp   | chalcopyrite        | BMS(1)                         |
| (Pt,Pd)S                              | cooperite, braggite, vysotskite                      | PGM(2)                         | Co-Pn | Co-rich pentlandite | BMS(2)                         |
| (Pt,Rh) <sub>2</sub> CuS <sub>4</sub> | cuprorhodsite, malanite                              | PGM(3)                         | Mlr   | millerite           | BMS(3)                         |
| (Ru, Ir, Os)S <sub>2</sub>            | erlichmannite, laurite                               | PGM(4)                         | Pn    | pentlandite         | BMS(4)                         |
| Alloys(Fe,Sn)                         | native platinum, rustenburgite, tetraferroplatinum   | PGM(5)                         | Py/Po | pyrite, pyrrhotite  | BMS(5)                         |
| Alloys(Sb,Bi)                         | geversite, insizwaite, stibiopalladinite, sudburyite | PGM(6)                         | Vio   | violarite           | BMS(6)                         |

Table 7 Modal mineralogy of investigated LG-6 and LG-6A drill core intersections. All values are presented as area%.

| Mineral              | ZK144-LG-6 | EL28D-LG-6 | ZK149-LG-6 | EL32D-LG-6A | ZK136D-LG-6A | SC42-LG-6A |
|----------------------|------------|------------|------------|-------------|--------------|------------|
| Alteration Silicates | 5.7        | 0.9        | 3.1        | 11.8        | 2.2          | 0.4        |
| Carbonates           | <0.1       | <0.1       | <0.1       | 3.7         | <0.1         | <0.1       |
| Chromite             | 83.3       | 89.9       | 93.5       | 66.8        | 68.2         | 75.2       |
| Silicates            | 10.8       | 9.2        | 3.4        | 17.7        | 29.6         | 24.4       |
| Others               | 0.2        | <0.1       | <0.1       | <0.1        | <0.1         | <0.1       |
| Total                | 100.0      | 100.0      | 100.0      | 100.0       | 100.0        | 100.0      |

Table 8 Summary of important parameters of determined assemblages for mineral beneficiation.

| Assemblage | PGM species                          | BMS species             | Mineral association/ Gangue mineralogy |                      | PGM grain sizes                        | BMS particle sizes        |
|------------|--------------------------------------|-------------------------|--|----------------------|--|---------------------------|
| (I)        | PGM-(Cu)-sulfides; alloys of Fe, Sn  | Pn + Ccp + Py + Mlr     | BMS, Chromite, Silicate                |                      | min: <1µm,<br>max: 20µm,<br>avg: 4-5µm | max: 300µm,<br>avg: >30µm |
| (II)       |                                      |                         | alteration silicates                   | silicates            |  |                           |
| (III)      | PGM-(sulf)arsenides; alloy of Sb, Bi | (Co-rich) Pn ± Ccp ± Py |  | carbonates, chromite |  |                           |

## Figures

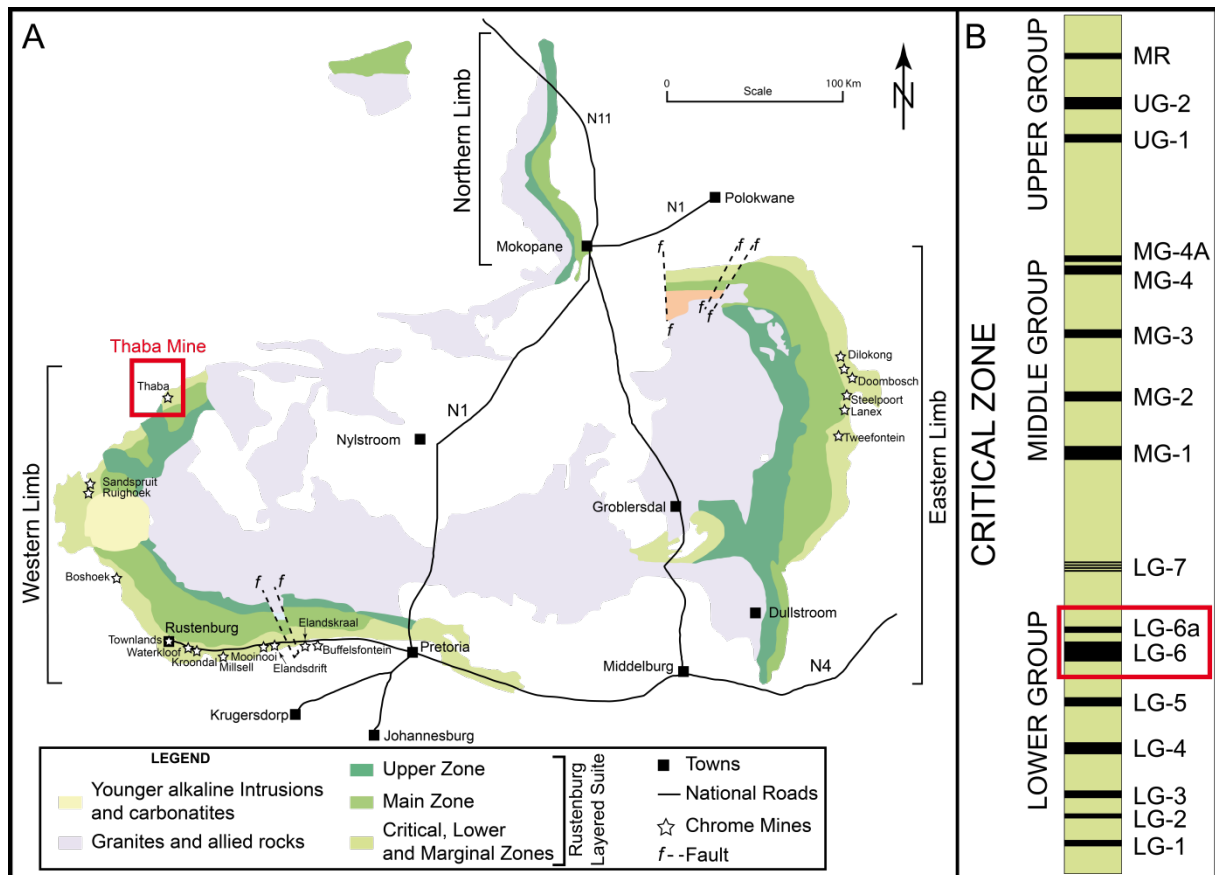
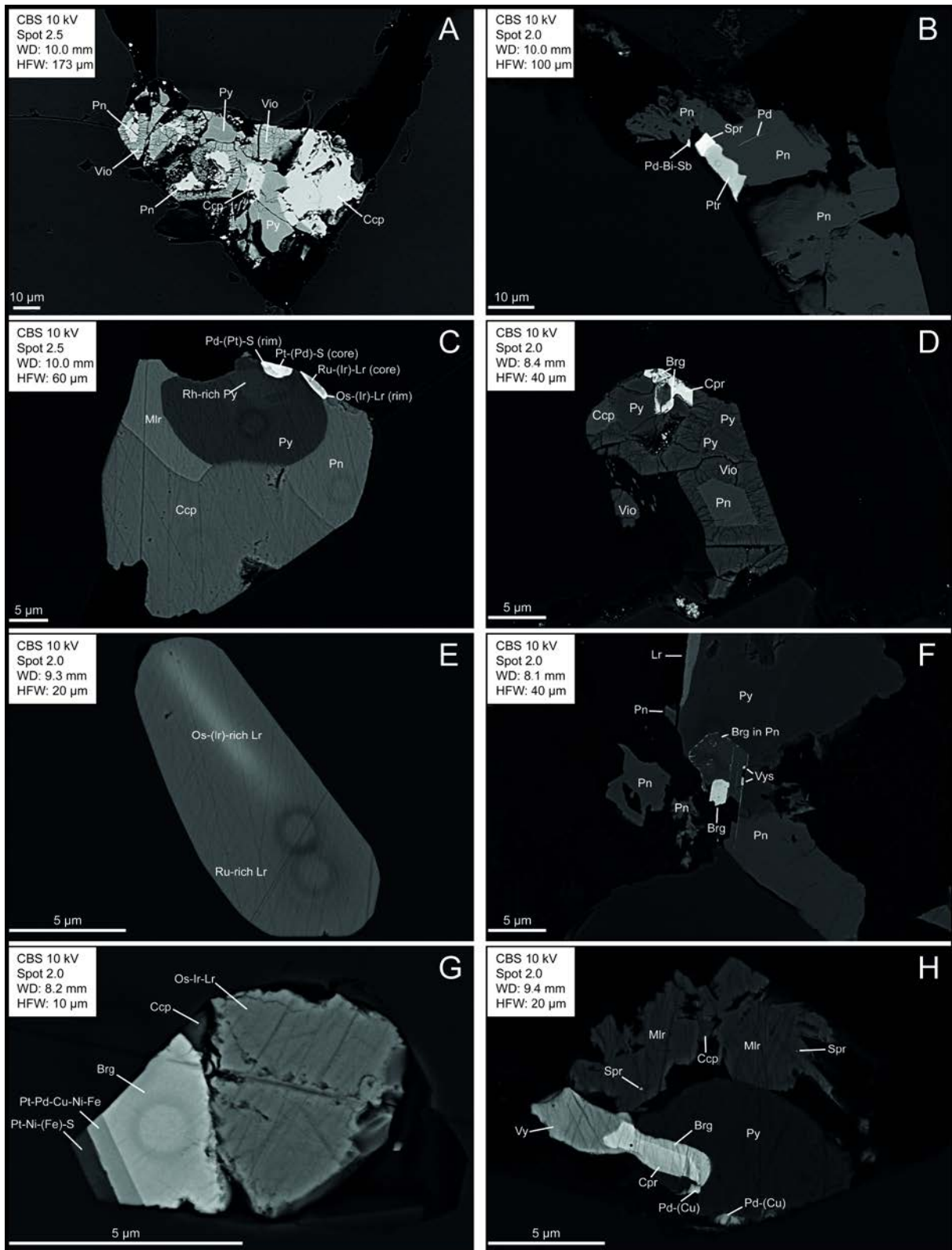


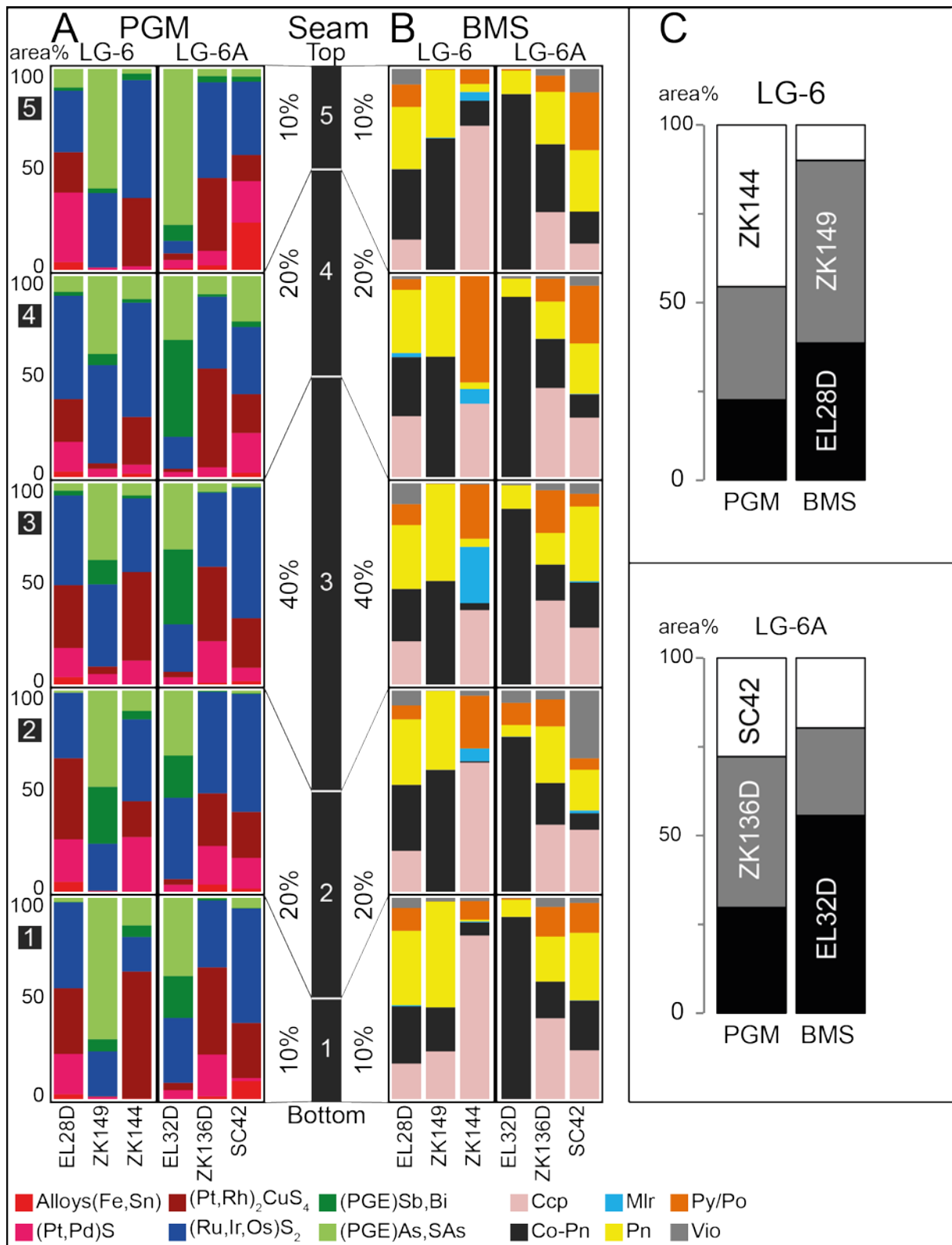
Figure 1(A) Geological map of the Bushveld Complex. The location of the Thaba mine is marked in red. (B) Stratigraphic column of the Critical Zone showing the positions of the major chromitite seams and of the Merensky Reef (MR) (modified from Oberthür et al. 2016).



1  
2 Figure 2 Back-scattered electron (BSE) images of sulfides and PGM in polished sections of the LG-6 and LG-6A  
3 chromitite seams. Scale bar and SEM conditions are displayed in the images. (A) Sulfide aggregate containing  
4 pyrite (Py), chalcopyrite (Ccp), pentlandite (Pn) and violarite (Vio) in intersection EL28D-LG-6. Please note the  
5 alteration of pentlandite into violarite. (B) Platarsite (Plr), Sperrylite (Spr) and a PdBiSb alloy intergrown with Pn  
6 in intersection ZK149-LG-6. Please note the Pd dominated lamellae in Pn. (C) Sulfide droplet interstitial to  
7 chromite containing millerite (Mlr), Py, Pn and Ccp in intersection SC42-LG-6A. Note the segregation of PPGE  
8 and IPGE into different discrete PGM (Braggite (Brg) and Laurite (Lr), respectively) as well as the exsolution of  
9 Rh into pyrite (slightly brighter). Both grains show further segregation of Pt and Ru, respectively, located in the  
10 core and Pd and Os in the rim.

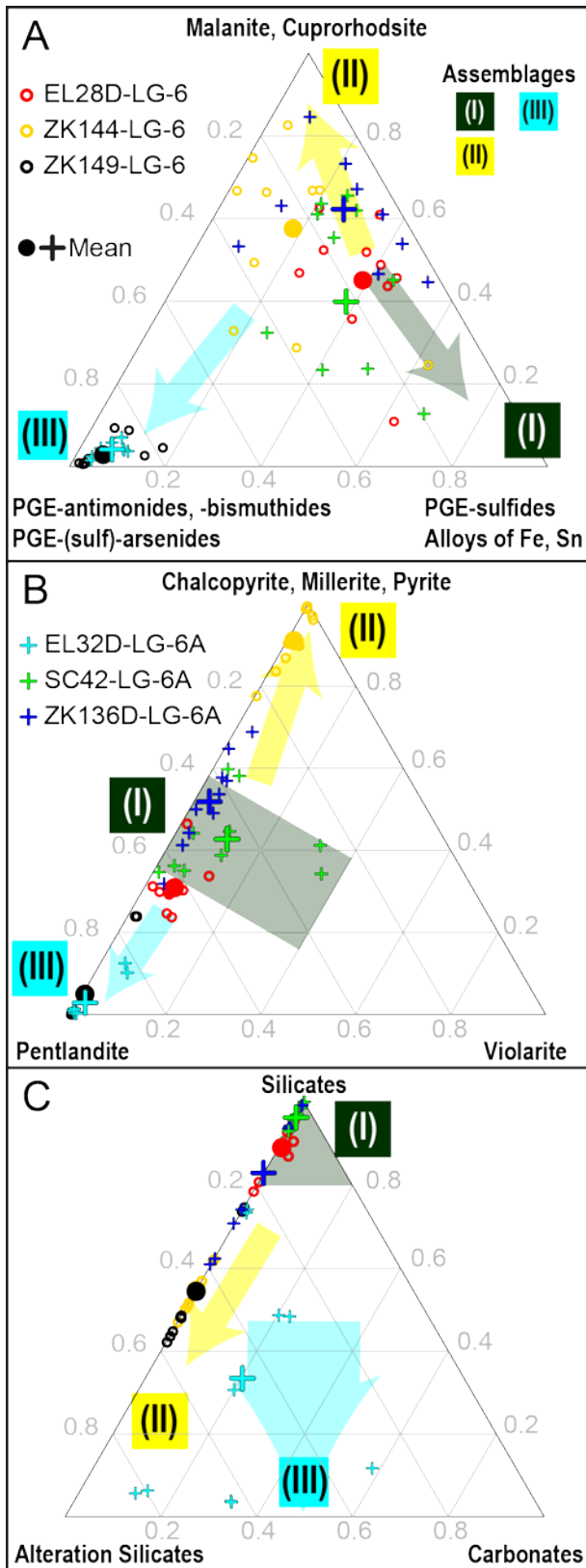


11 (Brg) – Cooperite (Cpr) compositions in intersection EL28D-LG-6. Note the alteration of Pn into Vio. (E) Laurite  
12 grain as an inclusion in chromite in intersection SC42-LG-6A. Note the brighter area due to a higher  
13 concentration of Os and Ir. (F) Brg intergrown with sulfides in intersection ZK136D-LG-6A. Note the exsolution  
14 of Pt-Pd into the surrounding sulfides during cooling – keeping the original shape of the Brg grain and the  
15 formation of Vysotskite (Vys). (G) PGE particle composed of various PGM, both, sulfides (Brg and Lr) as well as  
16 alloys as inclusion in chromite in intersection ZK136D-LG-6A. (H) Pt-Pd sulfide of various composition ranging  
17 from Vys to Cpr intergrown with various sulfides in intersection SC42-LG-6A. Note the nanoinclusions of Spr in  
18 Mlr and a Pd-Cu rich bright phase.  
19

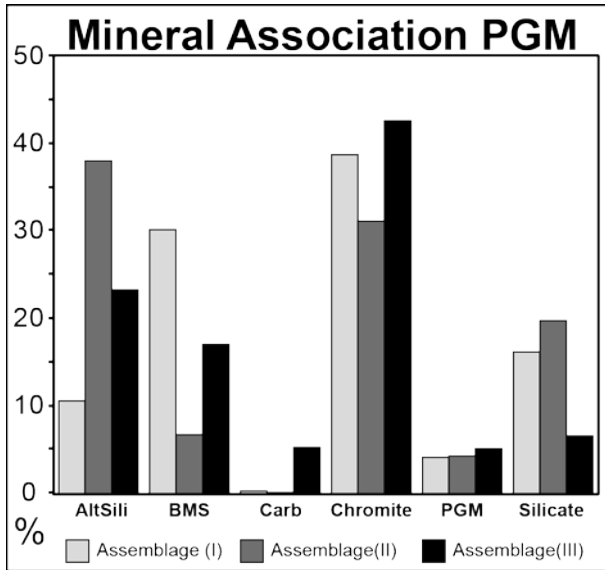


20  
 21 Figure 3 Comparison of MLA thin sections data for drill cores of the LG-6 and LG-6A seam intersections. Thirty  
 22 samples were analyzed for each seam. (A, B) 100% stacked histograms that show the PGM (for detailed  
 23 grouping see text) (A) and sulfide (B) assemblages of LG-6 and LG-6A intersections from bottom to top (sectors  
 24 1 to 5). (C) Based on normalized thin sections, PGM and BMS area within the sample ( $\mu\text{m}^2$ ) were summed for  
 25 each drill core separately. The relative distribution of PGM and BMS (area%) in seam LG-6 and LG-6A,  
 26 respectively, is shown in stacked bar plots. "Seam" displays the division into five sampled distinct sectors  
 27 (sectors 1-5, from bottom to top) and each sector was sampled randomly to prepare an individual polished thin  
 28 section for study. For details see text.

29



30  
 31 Figure 4 Modal proportions of all thin sections (and the corresponding means) of LG-6 and LG-6A samples for  
 32 (A) PGM, (B) BMS, and (C) Gangue mineral groups projected in a ternary diagram.  
 33



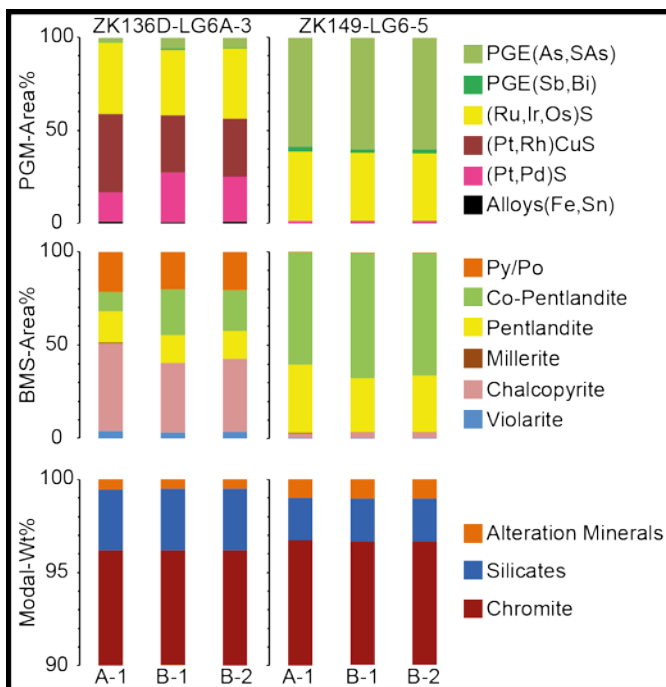
34  
 35 Figure 5 Histogram of mineral association of PGM comparing assemblages (I), (II), (III). Assemblage (I) displays  
 36 high abundances of PGM-sulfides and alloys of Fe and Sn, while assemblage (II) is dominated by PGM-Cu-  
 37 sulfides. Assemblage (III), on the other hand, comprises high amounts of PGM-sulfarsenides and PGM-  
 38 arsenides together with PGE-antimonides, -bismuthides and scant -bismuthotellurides.  
 39

40

41 **Appendix A**

42 *Methods – Reproducibility of MLA data*

43 Quantification of the reproducibility and statistical significance of the analyses was performed  
 44 through (1) analyzing a thin section twice in the same analytical run (in-run duplication), and (2) by  
 45 re-polishing of the sample to analyze a second surface of the same sample. Results are displayed in  
 46 Figure A1. In-run analysis displays a relative standard deviation ( $2\sigma$ ) for total area%<sup>PGM</sup> of 5 % for  
 47 sample ZK149-LG-6 and 4 % for sample ZK136D-LG-6A, while total area%<sup>BMS</sup> and wt%<sup>Modal</sup> show  $2\sigma$   
 48 values below 2 % and 0.5 % for both samples, respectively. The combined in-run duplicates of surface  
 49 A with surface B, 2SDs are 32 % in sample ZK149 and 15 % in sample ZK136D-LG-6A for total  
 50 area%<sup>PGM</sup>, while total area%<sup>BMS</sup> and total area%<sup>Gangue</sup> show  $2\sigma$  values below 2 % and 3 % for both  
 51 samples, respectively.



52 Figure A1 Stacked histograms showing the proportions of PGM species (in terms of area%<sup>PGM</sup>), base  
 53 metal sulfides (in terms of area%<sup>BMS</sup>), and the modal mineralogy in wt% for samples ZK136D-LG-6A-3  
 54 and ZK149-LG-6-5. A-1: surface A; B-1: surface B-1; B-2: in-run duplicate of surface B-1.  
 55

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58 *Methods - Electron probe microanalysis (EPMA)*

59 Quantitative analyses were also performed for amphibole, chromite, clinopyroxene, feldspar, olivine,  
60 orthopyroxene, serpentine and talc. Concentrations of Na, Mg, Al (all TAP); Si, K, Ca (all PETJ); Cr, V, Ti  
61 (all PETL); Zn, Ni, Co, Fe, Mn (all LIFH) were measured using  $K_{\alpha}$  – lines. Certified reference materials  
62 provided by ASTIMEX Standards Ltd. were used for calibration. An electron beam diameter of 5  $\mu\text{m}$   
63 was set at 20kV/ 12 nA and the ZAF approach was used for matrix correction (atomic number–  
64 absorption–fluorescence; Philibert, 1963; Reed, 1965; Philibert & Tixier, 1968). Dwell times were set  
65 to 15s (Na), 20s (Si, Cr, Ni), and 30s (K, Ca, Fe, Mn), 40s (Mg, Al, V, Ti, Zn, Co). Offline overlap  
66 corrections were performed for Zn  $L_{\beta 1}$  on Na  $K_{\alpha 1,2}$ ; Mn  $K_{\alpha 1}$ , Ti  $K_{\alpha 1}$ , Cr  $K_{\beta 1,3}$  on Al  $K_{\alpha 1,2}$ ; Ni  $K_{\alpha 1}$  on Ca  $K_{\alpha 1}$ ; V  
67  $K_{\beta 1,3}$  on Cr  $K_{\alpha 1}$ ; Ti  $K_{\beta 1,3}$  on V  $K_{\alpha 1}$ ; Mn  $K_{\beta 1,3}$  on Fe  $K_{\alpha 1}$ ; and Cr  $K_{\beta 1,3}$  on Mn  $K_{\alpha 1}$  as explained in Osbahr et al.  
68 (2015).

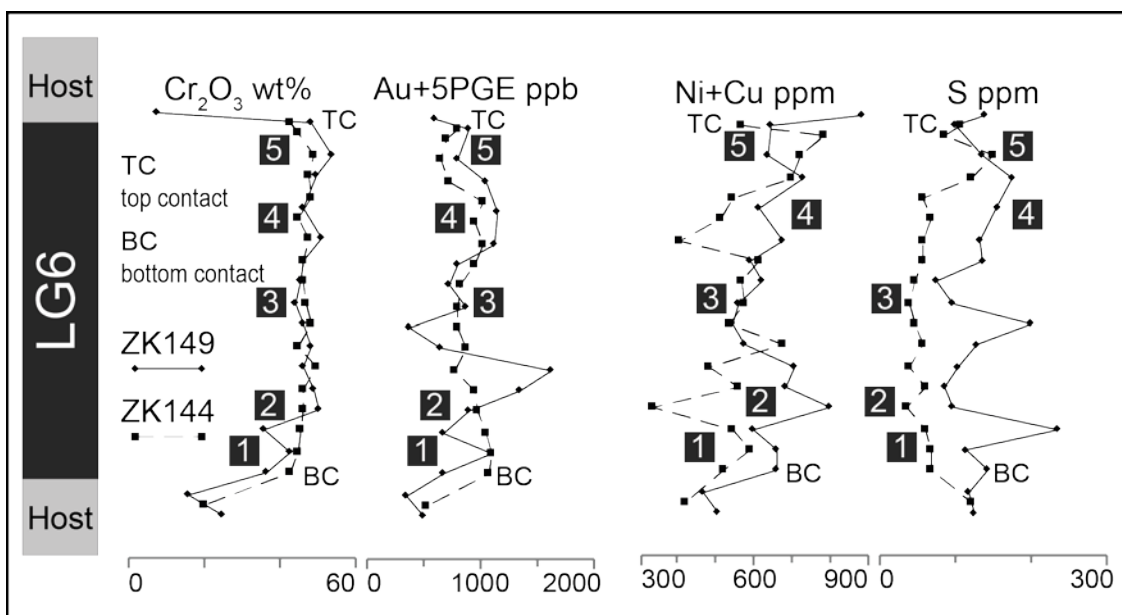
69 Finally, EPMA results were used to generate mineral standard spectra with a known composition for  
70 MLA measurements. Therefore, mineral standards were taken close to EPMA measurements. This  
71 approach allows a detailed differentiation of silicates, such as different feldspar compositions,  
72 amphiboles, pyroxenes and alteration silicates.

73 *Results - Geochemistry*

74 5PGE+Au (Pt,Pd,Rh,Ir,Ru,Au) contents are around 1 ppm for LG-6 samples and 1.5 ppm for the LG-6A.  
75 In general, the total sulfur content ranges between 100 – 200 ppm (<250 ppm), except intersection  
76 ZK144-LG-6, which contains only around 70 ppm. Detailed information are provided with the  
77 electronic supplementary material.

78 Figure A2 displays detailed geochemical assays of two selected drill core intersections of the LG-6  
79 (ZK149-LG-6 and ZK144-LG-6). The profiles display a rather stable  $\text{Cr}_2\text{O}_3$  content around 43-51 wt%.  
80 Slight fluctuations, especially in ZK149-LG-6 display massive chromitite intercalated with  $\sim 1$  cm  
81 pyroxene-rich bands or pyroxene oikocrysts. 5PGE+Au contents are 900 ppb and 950 ppb with a  
82 Pt/Pd ratio around 2.2 and 4 and Pt/Ru ratios around 0.9 for intersections ZK144-LG-6 and ZK149-LG-

83 6, respectively. Both profiles display 5PGE+Au profiles with a “M-shape”; ZK144-LG-6 displays a  
 84 smooth variation in the overall PGE contents, ranging from ~700 ppb to 1100 ppb, while ZK149-LG-6  
 85 displays spikes with minimum single element contents of roughly 370 ppb and a maximum of  
 86 1600 ppb. The Ni+Cu plot displays different patterns, however, with variable contents for both of the  
 87 examples. Cu/ (Cu+Ni) ratios are 0.02 and 0.01 for ZK144-LG-6 and ZK149, respectively. In general,  
 88 the LG-6 is sulfur poor, with contents ranging from 70 ppm (ZK144-LG-6) to 130 ppm (ZK149-LG-6),  
 89 on average.



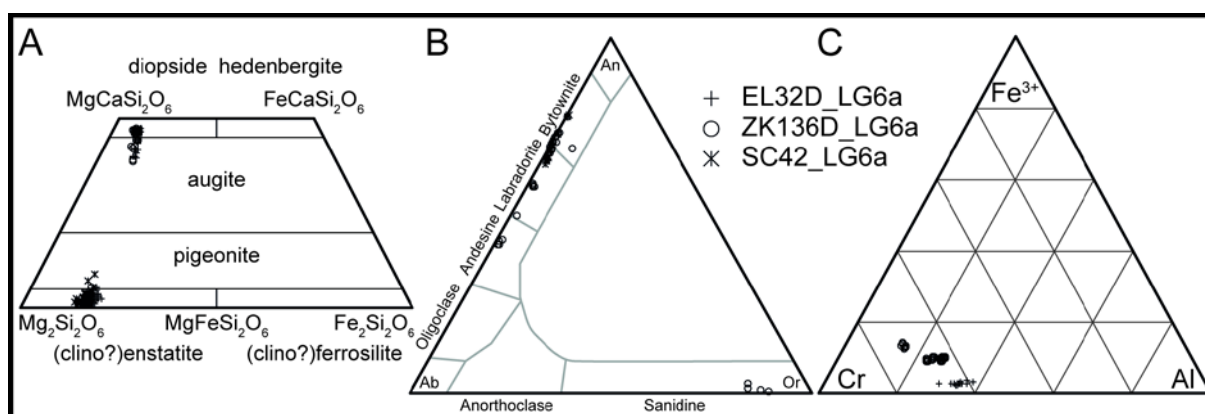
90  
 91 Figure A2 Detailed geochemistry of Cr<sub>2</sub>O<sub>3</sub>, Au+PGE (6E), Ni+Cu, and S concentrations versus  
 92 stratigraphic height, normalized to 100 (ZK149: 83 cm; ZK144: 93 cm) in the LG6. Note the bottom  
 93 (BC) and top contact (TC) of the intersections. 6E analyzed by Ni fire assay, aqua-regia digestion and  
 94 ICP-MS finish, sulfur analyzed by “LECO” combustion.

95 *Results - EPMA analyses – silicate minerals and chromite*

96 Altogether, 359 analyses on LG-6A samples were obtained to describe the mineral chemistry of  
 97 various silicate minerals as well as chromite in the LG seams. Table A1 displays representative  
 98 analyses of this sample set, extended by some analyses of important mineral compositions, namely  
 99 Al-rich chromite, olivine, and serpentine, contained in a MG sample set due to a lack of these  
 100 minerals in the measured LG sections. Mineral standard spectra for further MLA analysis were taken  
 101 on grains with a known composition for a better identification during quantitative mineralogical  
 102 assessment.

103 Figure A3A displays two different types of pyroxenes. One type yields an orthopyroxene  $X_{En}=78-85$ ,  
 104  $X_{Fs} = 13-19$  and  $X_{Wo} = 0.5-9$  and is hereafter referred to as enstatite. The second type shows a slight  
 105 trend from augitic to mainly diopsidic compositions and is composed of  $X_{En}=45-52$ ,  $X_{Fs} = 5-9$  and  $X_{Wo} =$   
 106  $38-48$  (hereafter referred to as diopside). Amphiboles can be best described as magnesio-  
 107 hornblende. Feldspars shown in Figure A3B yield plagioclase compositions ranging from andesine to  
 108 bytownite, however, also minor albite and anorthite were analyzed by EDX measurement.  
 109 Furthermore, sanidine with  $X_{Or}$  up to 90 % was encountered. For both, pyroxene and feldspar, there  
 110 were no significant differences between samples detected. Chromite was analyzed in sample  
 111 EL32D\_LG-6A and ZK136D\_LG-6A (Figure A3C). Data might indicate differences between samples, as  
 112 sample EL32D\_LG-6A displays low  $Fe^{3+}$  and slightly higher Al contents than ZK136D\_LG-6A.

113



114 Figure A3 Ternary plots of EPMA measurements. (A) Classification diagram of pyroxene after  
 115 Morimoto (1988). Both types of pyroxene – enstatite (orthopyroxene) and (augite)-diopside are  
 116 rather chemically homogeneous. (B) Classification diagram of feldspar with endmembers anorthite  
 117 ( $An$ ), albite ( $Ab$ ), alkali feldspar/ orthoclase ( $Or$ ). Beside plagioclase with variable composition alkali  
 118 feldspar with sanidine composition was detected. (C) Ternary plot of trivalent cations, Cr, Al and  $Fe^{3+}$   
 119 in chromite. Note that the data show only little variation, however, all analyses of intersection  
 120 EL32D\_LG-6A display a lower concentration in  $Fe^{3+}$  compensated by a higher Al content.

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127 Table A1 Representative EPMA analyses of all relevant silicate minerals and chromite, used for the  
 128 MLA Mineral Reference List.

| No.                              | 346      | 330         | 23       | 207     | 221     | 120     | 228         | 391        | 197       | 445     | 192       | 166    | 132        |
|----------------------------------|----------|-------------|----------|---------|---------|---------|-------------|------------|-----------|---------|-----------|--------|------------|
| Core                             | ZK136D   | EL32        | EL32D    | EL32D   | SC42    | ZK146   | SC42        | ZK136D     | EL32D     | EL32    | EL32D     | EL32D  | ZK146      |
| Seam                             | LG-6A    | MG1         | LG-6A    | LG-6A   | LG-6A   | MG1     | LG-6A       | LG-6A      | LG-6A     | MG1     | LG-6A     | LG-6A  | MG1        |
| Mineral                          | Chromite | Al-Chromite | Chromite | ClinoPx | OrthoPx | Olivine | Plagioclase | K-Feldspar | Muscovite | Biotite | Amphibole | Talc   | Serpentine |
| <i>all data in wt%</i>           |          |             |          |         |         |         |             |            |           |         |           |        |            |
| Na <sub>2</sub> O                | bdl      | bdl         | bdl      | 0.35    | 0.01    | 0.02    | 2.36        | 1.19       | 0.11      | 0.26    | 1.84      | 0.05   | 0.02       |
| MgO                              | 4.84     | 14.26       | 6.56     | 18.00   | 31.44   | 45.36   | 0.04        | 0.01       | 0.54      | 18.84   | 16.64     | 25.69  | 35.97      |
| Al <sub>2</sub> O <sub>3</sub>   | 7.43     | 48.11       | 17.40    | 2.00    | 1.38    | 0.00    | 32.54       | 18.33      | 34.88     | 20.71   | 10.25     | 0.38   | 0.01       |
| SiO <sub>2</sub>                 | 0.04     | 0.72        | 0.03     | 53.28   | 55.32   | 40.03   | 49.18       | 63.29      | 47.30     | 39.39   | 47.60     | 59.28  | 40.58      |
| K <sub>2</sub> O                 | bdl      | bdl         | bdl      | bdl     | 0.01    | 0.00    | 0.04        | 15.44      | 9.66      | 9.24    | 0.15      | 0.02   | 0.03       |
| CaO                              | bdl      | bdl         | bdl      | 19.44   | 0.74    | 0.03    | 15.64       | 0.10       | 0.26      | 1.52    | 11.70     | 0.20   | 0.03       |
| Cr <sub>2</sub> O <sub>3</sub>   | 50.81    | 14.86       | 46.53    | 0.99    | 0.56    | 0.00    | 0.23        | bdl        | 0.02      | 0.06    | 2.01      | 0.09   | 0.01       |
| V <sub>2</sub> O <sub>3</sub>    | 0.46     | 0.34        | 0.44     | 0.05    | 0.03    | 0.00    | 0.01        | 0.02       | 0.02      | 0.01    | 0.11      | 0.01   | 0.01       |
| TiO <sub>2</sub>                 | 0.47     | 0.05        | 0.84     | 0.22    | 0.09    | 0.00    | bdl         | 0.04       | bdl       | 0.27    | 1.30      | 0.02   | 0.02       |
| ZnO                              | 0.13     | 0.76        | 0.11     | bdl     | bdl     | bdl     | bdl         | bdl        | bdl       | bdl     | bdl       | bdl    | bdl        |
| NiO                              | 0.12     | 0.13        | 0.14     | 0.04    | 0.08    | 0.29    | bdl         | 0.04       | bdl       | 0.11    | 0.12      | 0.05   | 0.32       |
| CoO                              | 0.13     | 0.13        | 0.09     | 0.02    | 0.03    | 0.03    | bdl         | bdl        | 0.01      | 0.02    | 0.02      | 0.01   | 0.03       |
| MnO                              | 0.32     | 0.09        | 0.30     | 0.16    | 0.24    | 0.20    | bdl         | bdl        | 0.01      | 0.02    | 0.08      | 0.04   | 0.05       |
| FeO                              | 25.33    | 16.88       | 24.52    | 5.40    | 9.88    | 13.92   | 0.28        | 0.35       | 2.24      | 4.82    | 5.91      | 8.69   | 7.59       |
| Fe <sub>2</sub> O <sub>3</sub>   | 9.35     | 1.60        | 1.87     |         |         |         |             |            |           |         |           |        |            |
| Total                            | 99.42    | 97.92       | 98.82    | 99.97   | 99.82   | 99.89   | 100.31      | 98.81      | 95.05     | 95.27   | 97.72     | 94.54  | 84.68      |
| <i>all data is given as apfu</i> |          |             |          |         |         |         |             |            |           |         |           |        |            |
| Na                               |          |             |          | 0.025   | 0.001   | 0.001   | 0.208       | 0.108      | 0.015     | 0.035   | 0.198     | 0.014  |            |
| Na (A)                           |          |             |          |         |         |         |             |            |           |         | 0.315     |        |            |
| Mg                               | 2.019    | 4.763       | 2.591    | 0.979   | 1.653   | 1.693   | 0.003       | 0.001      | 0.053     | 1.981   | 3.565     | 5.096  | 5.302      |
| Al                               | 3.678    | 19.052      | 8.152    | 0.086   | 0.058   |         | 1.748       | 1.014      | 2.731     | 1.722   | 1.735     | 0.060  | 0.002      |
| Al IV                            |          |             |          | 0.055   | 0.049   |         |             |            | 0.857     | 1.221   | 1.160     | 0.060  |            |
| Al VI                            |          |             |          | 0.031   | 0.008   |         |             |            | 1.874     | 0.502   | 0.576     |        | 0.002      |
| Si                               | 0.024    | 0.323       | 0.014    | 1.945   | 1.951   | 1.002   | 2.241       | 2.969      | 3.143     | 2.779   | 6.840     | 7.889  | 4.012      |
| Si (T1)                          |          |             |          |         |         |         |             |            |           |         | 2.840     |        |            |
| K                                |          |             |          |         |         |         | 0.003       | 0.924      | 0.819     | 0.831   | 0.027     | 0.004  | 0.004      |
| Ca                               |          |             |          | 0.760   | 0.028   | 0.001   | 0.763       | 0.005      | 0.018     | 0.115   | 1.802     | 0.028  | 0.004      |
| Cr                               | 16.873   | 3.948       | 14.627   | 0.029   | 0.016   |         | 0.008       | 0.000      | 0.001     | 0.004   | 0.229     | 0.010  | 0.001      |
| V                                | 0.160    | 0.095       | 0.146    | 0.002   | 0.001   |         |             | 0.001      | 0.001     | 0.001   | 0.014     | 0.001  | 0.001      |
| Ti                               | 0.199    | 0.016       | 0.333    | 0.006   | 0.002   |         |             | 0.001      |           | 0.014   | 0.140     | 0.002  | 0.001      |
| Zn                               | 0.026    | 0.126       | 0.022    |         |         |         |             |            |           |         |           |        |            |
| Ni                               | 0.026    | 0.024       | 0.029    | 0.001   | 0.002   | 0.006   |             | 0.002      |           | 0.006   | 0.014     | 0.005  | 0.025      |
| Co                               | 0.029    | 0.022       | 0.019    | 0.001   | 0.001   | 0.001   |             |            | 0.001     | 0.001   | 0.003     | 0.001  | 0.002      |
| Mn                               | 0.076    | 0.017       | 0.068    | 0.005   | 0.007   | 0.004   |             |            | 0.001     | 0.001   | 0.009     | 0.004  | 0.004      |
| Fe <sup>2+</sup>                 | 5.928    | 3.163       | 5.433    | 0.165   | 0.291   | 0.292   | 0.011       | 0.014      | 0.124     | 0.284   | 0.710     | 0.967  | 0.628      |
| Fe <sup>3+</sup>                 | 2.955    | 0.404       | 0.560    |         |         |         |             |            |           |         |           |        |            |
| Vac                              |          |             |          |         |         |         |             |            |           |         | 0.658     |        |            |
| Total                            | 31.993   | 31.952      | 31.995   | 4.003   | 4.010   | 3.000   | 4.985       | 5.038      | 6.907     | 7.776   | 15.600    | 14.082 | 9.985      |
| Cr/Al                            | 4.587    | 0.207       | 1.794    |         |         |         |             |            |           |         |           |        |            |
| Mg#                              | 0.254    | 0.601       | 0.323    | 0.86    | 0.85    |         |             |            |           |         |           |        |            |
| FFE                              | 0.333    | 0.113       | 0.093    |         |         |         |             |            |           |         |           |        |            |
| Cr#                              | 0.821    | 0.172       | 0.642    |         |         |         |             |            |           |         |           |        |            |
| Cr/Fe                            | 1.33     | 0.71        | 1.56     |         |         |         |             |            |           |         |           |        |            |
| Sum IV                           |          |             |          | 2.000   | 2.000   |         |             |            |           |         |           |        |            |
| Sum VI                           |          |             |          | 2.003   | 2.010   |         |             |            |           |         |           |        |            |
| X <sub>Wo</sub>                  |          |             |          | 39.92   | 1.43    |         |             |            |           |         |           |        |            |
| X <sub>En</sub>                  |          |             |          | 51.43   | 83.81   |         |             |            |           |         |           |        |            |
| X <sub>Fs</sub>                  |          |             |          | 8.65    | 14.77   |         |             |            |           |         |           |        |            |
| Fo                               |          |             |          |         |         | 85.10   |             |            |           |         |           |        |            |
| An                               |          |             |          |         |         |         | 78.35       | 0.50       |           |         |           |        |            |
| Or                               |          |             |          |         |         |         | 0.26        | 89.08      |           |         |           |        |            |
| X <sub>Mg</sub>                  |          |             |          |         |         |         |             |            | 0.053     | 0.637   | 0.834     |        |            |
| Na+K(A)                          |          |             |          |         |         |         |             |            |           |         | 0.342     |        |            |

129 Please note the analyses for Al-chromite, olivine, biotite and serpentine were taken from a different  
 130 sample set of MG1 samples from the same locality. Calculation of chromite is normalized to 32 O;  
 131 pyroxene is normalized to 6 O; olivine is normalized to 4 O; feldspar is normalized to 8 O; mica is  
 132 normalized to 11 O; amphibole is normalized to 23 O; talc is normalized to 22 O; and serpentine is  
 133 normalized to 14 O. apfu – atoms per formula unit.

134

135 ANOVA

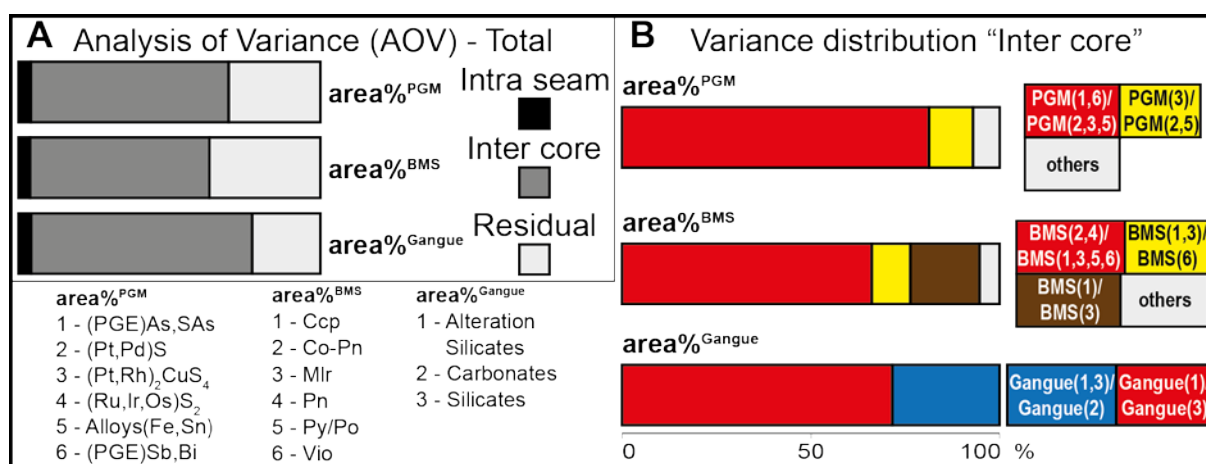
136 An ANOVA was performed – followed by standard F-tests – to investigate the variability of the  
137 samples and to determine assemblages for classification according to the sub-compositions  
138  $\text{area}\%^{\text{PGM}}$ ,  $\text{area}\%^{\text{BMS}}$  and  $\text{area}\%^{\text{Gangue}}$  (see legend in Figure A4A). Variability of the samples was tested  
139 for four explanatory factors (“surfaces A/B”, “intra seam”, “inter core” and “inter seam” (see in the  
140 methods section for definition of these explanatory factors). “Surfaces A/B” compares the results  
141 obtained for the two different polished surfaces (A and B) analysed for each sample. This explanatory  
142 factor is found to have no significant influence on the tested ANOVA model, which allows the  
143 conclusion that the exact position of the sample surface examined by MLA has no significant effect  
144 on the results obtained. Most notably, it suggests that the “nugget effect” (small scale variability) is  
145 negligible.

146 The explanatory factor “intra seam” considers the variance between samples of a different  
147 stratigraphic position within a sampled seam, *i.e.* if seams have systematically different mineral  
148 assemblages within their respective confines. The factor “intra seam” does have a significant  
149 influence for the variance of both,  $\text{area}\%^{\text{PGM}}$  and  $\text{area}\%^{\text{BMS}}$  and a highly significant influence for the  
150 variance of  $\text{area}\%^{\text{Gangue}}$ . The two factors “inter seam” (considering systematic differences between  
151 the two studied seams, LG-6 and LG-6A) and “inter core” (considering systematic differences  
152 between different drill core intersections) show close interrelation, meaning that variance induced  
153 by “inter seam” will be a part of the variance induced by “inter core”. While the F-tests for the factor  
154 “inter seam” are non-significant for  $\text{area}\%^{\text{PGM}}$  and  $\text{area}\%^{\text{BMS}}$ , and highly significant for  $\text{area}\%^{\text{Gangue}}$ , the  
155 factor “inter core” yields highly significant results in all cases. This suggests that systematic  
156 differences between distinct seams intersected within the same drill core are of minor importance  
157 compared to differences encountered between drill cores from different sites on the mine lease  
158 area.

159 Figure 4A shows the variance distribution for the two most relevant factors. “Inter core” is by far the  
160 most important explanatory factor, explaining between 59 and 77 % of the total variance. In  
161 comparison, the factor “intra core” explains only 6 % of total variance – a value that is similar for all  
162 sub-compositions. As mentioned above, the factor “seam” was not considered further, as it explains  
163 only between 2 and 13 % of the “inter core” variance for  $\text{area}\%^{\text{PGM}}$  and  $\text{area}\%^{\text{BMS}}$  and  $\text{area}\%^{\text{Gangue}}$ . The  
164 factor “residuals” reports the variance that cannot be assigned to any of the previously mentioned  
165 factors and ranges between 23 and 35 % of the total variance. A detailed evaluation of the  
166 explanatory factor “inter core” for all sub-compositions based on ratios of certain mineral groups is  
167 shown in Figure A4B. The chosen ratios explain the bulk total variance, *i.e.* reflect the distribution of  
168 the group members. According to the ANOVA model the main part of the total variance in  $\text{area}\%^{\text{PGM}}$   
169 is invested in predicting the ratios [PGE -sulfarsenides and PGE-antimonides, -bismuthides vs. PGE –  
170 sulfides, -Cu-sulfides and -alloys of Fe,Sn]; *i.e.* these groups show a strong negative correlation and  
171 form distinctly different mineral assemblages. Furthermore, the ratio [PGE-Cu-sulfides vs. PGE –  
172 sulfides and –alloys of Fe,Sn] explains a significant portion of the total variance, and may be used to  
173 distinguish two more PGM mineral assemblages. The variability of IPGE -sulfides is rather low and  
174 only marginally explained by the factor “inter core”. The  $\text{area}\%^{\text{BMS}}$  “inter core” variance is mostly  
175 related to the variability of (Co-rich) pentlandite with respect to the geometric average of the other  
176 four BMS considered (chalcopyrite, millerite, pyrite and violarite). The ratios [chalcopyrite vs.  
177 millerite] and [chalcopyrite and millerite vs. violarite] do also have relevant contributions. The  
178 mentioned relationships correspond well to the  $\text{area}\%^{\text{PGM}}$  distribution, *i.e.* PGE-sulfarsenide/ -alloy of  
179 Sb,Bi-rich assemblages represent the same set of samples as the (Co-rich) pentlandite dominated  
180 BMS assemblages (forming assemblage (III)). On the other hand, PGE-(Cu)–sulfides and -alloys of  
181 Fe,Sn occur together with rather chalcopyrite, millerite, pyrite, violarite dominated samples –  
182 resulting in assemblage (I) and (II). The latter assemblages can be further subdivided into PGE-Cu-  
183 sulfides + chalcopyrite and millerite-rich samples (assemblage (II)), in contrast to assemblage (I)

184 where a PGE-sulfide-rich assemblage is associated with a rather variable BMS assemblage consisting  
 185 of chalcopyrite, pentlandite, pyrite and variable amounts of violarite.

186 In the  $area\%^{Gangue}$  sub-composition total “inter core” variance is best described using the ratios  
 187 [alteration silicates and silicates vs. carbonate] followed by [alteration silicates vs. silicates]. While  
 188 carbonate-rich samples are associated with PGM-BMS assemblage (III), silicate-rich samples  
 189 dominate PGM-BMS assemblage (I). Significant amounts of alteration silicates may occur in samples  
 190 belonging to PGM-BMS assemblages (II) and (III).



191 Figure A4 Stacked histograms of the results of an analysis of variance (AOV) of the modal mineralogy  
 192 for PGM, BMS and rock forming minerals (Others). (A) Total variance distribution for  $area\%^{PGM}$ ,  
 193  $area\%^{BMS}$  and  $area\%^{Gangue}$ . (B) Variance distribution based on defined ratios of certain  
 194 PGM/BMS/Gangue groups. For simplification, further reference to any group members will be as  
 195 following, e.g. (PGE)As,AsS = PGM(1); Alloys(Sb,Bi) = PGM(6); Ccp = BMS(1), etc. (cf. Table 6; legend  
 196 in (A)).  
 197  
 198

### 199 Cluster analysis

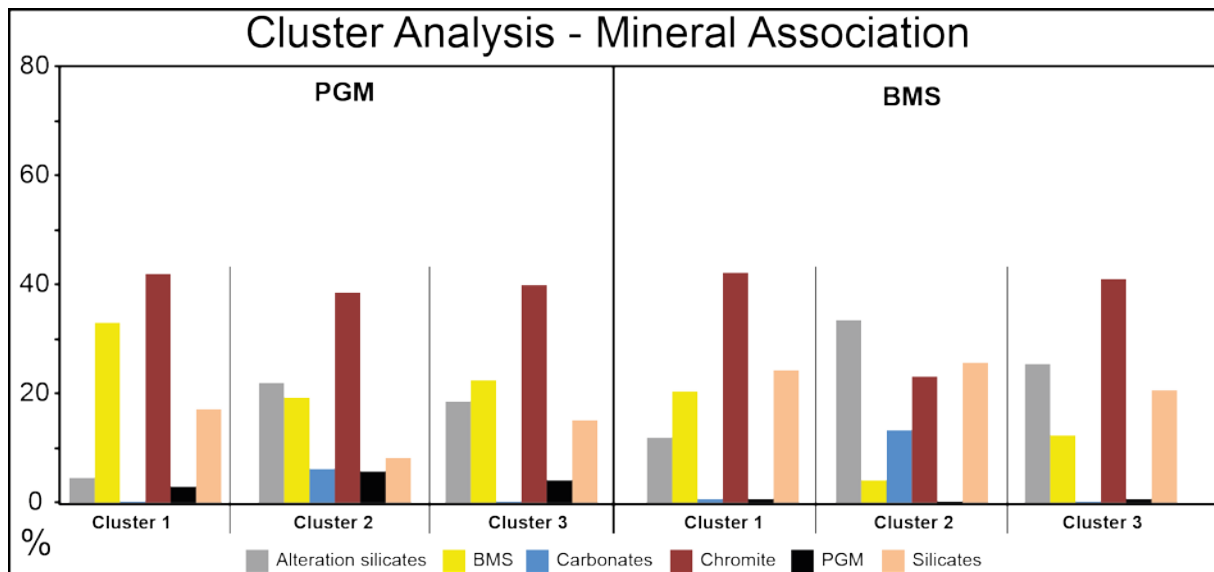
200 For a quantitative assessment of the variability of the mineral association a cluster analysis was  
 201 performed according to the sub-compositions of PGM (without IPGE sulfides) and BMS. The results  
 202 were linked to the mineral assemblages defined in the previous chapter (ANOVA).

203 Despite the very low total abundance of BMS (< 0.02 area% in all samples), PGM show a strong  
 204 preferred association with BMS; they also show a close association with alteration silicates, occurring  
 205 both in interstitial positions and as inclusion. PGM are only to a much lesser extent associated with

206 silicates and chromite. PGM occur predominantly interstitial to chromite (with minor inclusions cf.  
207 Figure 2G). BMS show a very similar association to PGM.

208 The cluster analysis results in three distinct clusters: a dominant and two minor, for both PGM  
209 [ $n_{\text{Cluster1}} = 5$ ,  $n_{\text{Cluster2}} = 16$ ,  $n_{\text{Cluster3}} = 39$ ] and for BMS [ $n_{\text{Cluster1}} = 39$ ,  $n_{\text{Cluster2}} = 12$ ,  $n_{\text{Cluster3}} = 9$ ] (Figure A5).

210 PGM cluster 1 consists of samples with a high silicate and lower chromite content, where most PGM  
211 occur in association with BMS. Cluster 2 includes carbonate-rich samples and PGM associated with  
212 alteration silicates and carbonates. Cluster 3, on the other hand, represents high chromite, high  
213 alteration silicate and low silicate content samples; PGM in this cluster are intergrown with alteration  
214 silicates and silicates. BMS cluster 1 is marked by high silicate/ alteration silicate ratios, where BMS  
215 occur predominantly as polymineralic aggregates intergrown with both alteration silicates and  
216 silicates. In contrast, the BMS cluster 3 lacks BMS aggregates and includes samples with low  
217 silicate/alteration silicate ratios. BMS are closely associated with silicates and alteration silicates in  
218 these samples. Similar to PGM cluster 2, the BMS cluster 2 represents carbonate-rich samples. Both,  
219 PGM and BMS mineral associations thus define similar clusters - implying a separation of silicate-  
220 dominated from alteration silicate and/or carbonate-dominated clusters. BMS and PGM tend to be  
221 associated with alteration minerals, if present, hence, establishing cluster 1 as typical for assemblage  
222 (I) seems valid. Furthermore, cluster 2 can be assigned to assemblage (III), while cluster 3 mainly  
223 represents the alteration silicate-rich members of assemblage (II) and (III).



224 Figure A5 Histograms of average mineral associations of clusters defined by two cluster analysis for  
 225 sub-compositions PGM (without IPGE –sulfides) and BMS, respectively. Color codes are explained  
 226 below corresponding histograms. Cluster (I) displays high abundances of PGM-sulfides and alloys of  
 227 Fe and Sn together with BMS dominated by pentlandite + chalcopyrite + pyrite. Cluster (II) comprises  
 228 high amounts of PGM-sulfarsenides and PGM-arsenides together with PGE-antimonides, -  
 229 bismuthides and scant –bismuthotellurides. Corresponding BMS are strongly dominated by  
 230 pentlandite and Co-rich pentlandite. On the one hand, cluster (III) represents samples dominated by  
 231 PGM-Cu-sulfide corresponding with high amounts of chalcopyrite. On the other hand, a minor  
 232 amount of samples contains significant portions of PGM-sulfarsenides, -arsenides, -antimonides, and  
 233 -bismuthides corresponding with a high abundance of pentlandite.  
 234  
 235

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